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**General, Organic & Biological Chemistry, 5e (Timberlake)
Chapter 8 Gases**

8.1 Multiple-Choice Questions

1) Which of the following is NOT part of the kinetic theory of gases?

- A) A gas is composed of very small particles.
- B) There is very little empty space in a gas.
- C) Gas particles move rapidly.
- D) Gas particles do not attract or repel one another.
- E) Gas particles move faster when the temperature increases.

Answer: B

Objective: 8.1

Global Outcomes: GO2

2) In the kinetic molecular theory of gas behavior, the assumption is made that gas molecules

- A) move rapidly in random directions.
- B) are attracted to each other by strong forces.
- C) are close together in their container.
- D) move with a kinetic energy equal to their centigrade temperature.
- E) occasionally come to rest.

Answer: A

Objective: 8.1

Global Outcomes: GO2

3) According to the kinetic theory of gases, a gas can be compressed much more than a liquid or solid because

- A) a gas is composed of very small particles.
- B) the particles of a gas are very far apart.
- C) gas particles move rapidly.
- D) gas particles do not attract or repel one another.
- E) gas particles move faster when the temperature increases.

Answer: B

Objective: 8.1

Global Outcomes: GO2

4) According to the kinetic theory of gases, particles of a gas

- A) are very large.

Organic And Biological Chemistry Timberlake Test

Karen Timberlake



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