

Chapter 1: Intro to earth science

- **Geology**- The study of earth
- **Oceanography**- study of the ocean
- **Meteorology**- Study of the atmosphere and weather
- **Astronomy**- Study of the universe

1. What science is not used within the Earth sciences? **Chemistry, Bio, Physics, and mathematics are all used**
2. **Scientific method**: Making observation/collect data, develop a hypothesis, conduct experiment, Re-examining the hypothesis and accept or reject it, Make conclusion
3. **Hypothesis**: Untested explanation
4. **Theory**- Tested and confirmed hypothesis
A THEORY is a well-tested and widely accepted view that best explains certain scientific observations
5. **Paradigm**- A theory that explains a large number of interrelated aspects of the natural world
6. **Hydrosphere**- Includes all water on earth (fresh water found in streams, lakes, and glaciers)
7. **Atmosphere**: The envelope of gases surrounding the earth, thin blanket of air, over 3.5miles
8. **Biosphere**- includes all life
9. **Geosphere**- Solid earth, extends from the surface to the center of the planet. **LARGEST** of Earth's 4 spheres, consists of the crust, mantle, and core
 - **Crust**- thin, rocky outer skin. Less dense, many rock types
 - **Oceanic crust**- dense, basaltic rock
 - **Mantle**- more than 82% of earth's volume.
 - **Upper mantle**- Lithosphere: entire crust and uppermost mantle.
Asthenosphere: soft layer
 - **Lower mantle**- **VERY HOT** strengthens with depth
 - **Outer Core**: Liquid layer, earth's magnetic field
 - **Inner core**- solid
10. Soil belongs to the Geosphere, hydrosphere, atmosphere, and biosphere
11. The Crust is the thinnest layer of the earth
12. What is the source of the energy that powers the earth system? **The sun and heat from earth's interior**

Chapter 2: Matter and minerals

- **Definition of a mineral**: Natural, inorganic, solid, possess an orderly internal structure of atoms, have a definite chemical composition
- **Rock**- and naturally occurring solid mass of mineral or mineral like matter
- **Atoms** smallest particles of matter
 - (1) **Nucleus**- central part of an atom that contains, protons +, neutrons neutral charge
 - (2) **Electron shells**- surrounds nucleus, contains electrons - charges

Science Chapter Study Guide

The Princeton Review



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