

1. Identify the phrases that generally apply to molecular compounds,

- | | |
|---|--|
| <input type="checkbox"/> A c. Use covalent bonds | <input type="checkbox"/> B a. Are made of non-metals chemically joined together. |
| <input type="checkbox"/> C d. Use ionic bonds | <input type="checkbox"/> D e. Have high boiling points |
| <input type="checkbox"/> E b. Have low melting points | |

2. The attraction of the free-floating valence electrons for the positively charged metal ions.

- | | |
|---|--|
| <input type="checkbox"/> A Ionic bond | <input type="checkbox"/> B Metallic Bond |
| <input type="checkbox"/> C London dispersion forces | <input type="checkbox"/> D Covalent Bond |

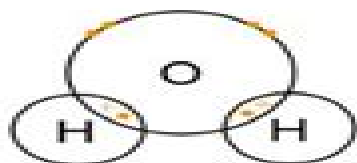
3. Which of the following elements will form a similar ion?

- | | |
|---------------------------------------|--------------------------------------|
| <input type="checkbox"/> A Be, Ca, Ba | <input type="checkbox"/> B Na, K, Li |
| <input type="checkbox"/> C Cl, Ar, Br | <input type="checkbox"/> D F, Cl, Br |

4. Which of the following is true about ionic bonding

- | | |
|---|---|
| <input type="checkbox"/> A It is weaker than the covalent bond | <input type="checkbox"/> B Electrons are donated from one element to another |
| <input type="checkbox"/> C Is made by the attraction between a cation and an anion. | <input type="checkbox"/> D It is mostly formed between elements from groups IA, IIA, IIIA and a nonmetal. |

5. What type of bond is formed between the hydrogen and oxygen atoms in this image?



- | | |
|--|---|
| <input type="checkbox"/> A Metallic bond | <input type="checkbox"/> B London Dispersion Forces |
|--|---|

Science Chapter 22 Chemical Bonds Worksheets

Richard Bailey

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Science Chapter 22 Chemical Bonds Worksheets Book Review: Unveiling the Power of Words

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