

Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

| | | | | | | | |
|----------|--------------------|---|-----------------|----------|---|-------|------------------------|
| Example: | NH_4^{+1} | & | I^{-1} | Formula: | <u>NH_4I</u> | Name: | <u>Ammonium Iodide</u> |
| 1. | NH_4 | & | O | Formula: | _____ | Name: | _____ |
| 2. | Ca | & | CO_3 | Formula: | _____ | Name: | _____ |
| 3. | HCO_3 | & | Na | Formula: | _____ | Name: | _____ |
| 4. | NO_3 | & | Mg | Formula: | _____ | Name: | _____ |
| 5. | K | & | SO_4 | Formula: | _____ | Name: | _____ |
| 6. | Li | & | PO_4 | Formula: | _____ | Name: | _____ |

Finding the Formula from the Name

| | | | | | |
|----------|---------------------|-------|--|----------|---------------------------------|
| Example: | Lithium Chloride | Ions: | <u>Li^{+1} & Cl^{-1}</u> | Formula: | <u>LiCl</u> |
| 1. | Sodium Oxide | Ions: | _____ | Formula: | _____ |
| 2. | Potassium Bromide | Ions: | _____ | Formula: | _____ |
| 3. | Calcium Nitride | Ions: | _____ | Formula: | _____ |
| 4. | Beryllium Sulfide | Ions: | _____ | Formula: | _____ |
| 5. | Magnesium Iodide | Ions: | _____ | Formula: | _____ |
| 6. | Aluminum Oxide | Ions: | _____ | Formula: | _____ |
| 7. | Ammonium Bromide | Ions: | _____ | Formula: | _____ |
| 8. | Calcium Bicarbonate | Ions: | _____ | Formula: | _____ |
| 9. | Potassium Phosphate | Ions: | _____ | Formula: | _____ |
| 10. | Aluminum Sulfate | Ions: | _____ | Formula: | _____ |

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