

Unit 1 Progress Check: MCQ

1. Which of the following numerical expressions gives the number of moles in 5.0 g of CaO ?

(A) $5.0 \text{ g} \times 56 \text{ g/mol}$

(B) $\frac{5.0 \text{ g}}{56 \text{ g/mol}}$

(C) $\frac{56 \text{ g/mol}}{5.0 \text{ g}}$

(D) $\frac{1}{5.0 \text{ g}} \times \frac{1}{56 \text{ g/mol}}$



2. In a lab a student is given a 21 g sample of pure Cu metal. Which of the following pieces of information is most useful for determining the number of Cu atoms in the sample? Assume that the pressure and temperature in the lab are 1.0 atm and 25°C .

(A) The molar mass of Cu

(B) The density of Cu at 25°C

(C) The volume of the Cu sample

(D) The ratio of the two main isotopes found in pure Cu



3. A 1.0 mol sample of which of the following compounds has the greatest mass?



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Reinhard Renneberg



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Sinha, Matthew Tontono, 2014-03-14 From the groundbreaking partnership of W H Freeman and Scientific American comes this one of a kind introduction to the science of biology and its impact on the way we live In *Biology for a Changing World* two experienced educators and a science journalist explore the core ideas of biology through a series of chapters written and illustrated in the style of a Scientific American article Chapters don't just feature compelling stories of real people each chapter is a newsworthy story that serves as a context for covering the standard curriculum for the non majors biology course Updated throughout the new edition offers new stories additional physiology chapters a new Electronic Teachers Edition and new pedagogy

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