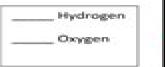
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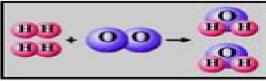
I. Chemical Changes in Matter

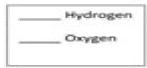
- A. Chemical Reaction
 - A change in which one or more substances are converted to ______ substances.



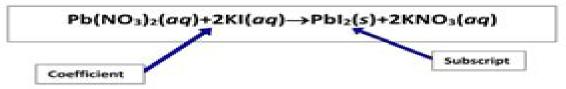
- B. Law of Conservation of Matter
 - In a chemical reaction, matter is not or ...
 - Atoms can only be ______.
 - 3. Discovered by Lavoisier.







C. Chemical Equations



- # of units of each substance.
- 2. Individual atom = atom

2Mg \Rightarrow 2 _____ of magnesium

3. Covalent substance = molecule

3CO₂ \Rightarrow 3 _____ of carbon dioxide

4. Ionic substance = unit

4MgO ⇒ 4 ______ of magnesium oxide

SYMBOL	MEANING
	produces, forms
•	plus, and
(s)	solid
Ø.	liquid
(9)	gas
(aq)	aqueous (solid dissolved in water)
Δ.	the reactants are heated

Science Chemical Reactions Note Taking Guide Answers

Eileen Ramsden

Science Chemical Reactions Note Taking Guide Answers:

Whispering the Strategies of Language: An Psychological Quest through **Science Chemical Reactions Note Taking Guide Answers**

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