#### Polyatomic lons Worksheet

#### Polyatomic Practice

LARGE CONTROL - C. C. MANUFACTOR CO.		
Name or write the formula for the following	ng polyatomic iona	
sulfate	CO <sub>2</sub> ®	
nitrite	MrvO <sub>3</sub> <sup>1</sup>	
perphosphate	SO <sub>3</sub> P	
hypoiodite	BrO <sub>8</sub> <sup>1</sup>	
chiloritie	GO <sub>4</sub> P	
phosphile	PO <sub>3</sub> 5	
percarbonate	CIO*	
bromate	1O <sub>2</sub> <sup>1</sup>	
hyposuffite	POv <sup>b</sup>	
permanganate	NO <sub>2</sub> <sup>1</sup>	
carbonite	50 <sub>4</sub> P	
Name or write the formula for the following	ng Type I polyatomic ionic compounds	
beryllium hydroxide	BaciOsla	
sodium nitrite	GiarGNOtis	
ammonium chloride	Agr8Or	
paloium bisulfate	MyCO	
as declarity one was a single-position.	NULL NUCL	
strantium sulfite	Alt(C <sub>0</sub> H <sub>2</sub> O <sub>0</sub> ) <sub>2</sub>	
aluminum acetate	SrSO <sub>6</sub>	
ammonium nitrate	FISCIO <sub>9</sub>	
magnesium hypocarbonite	Ca(HSO <sub>4</sub> ) <sub>8</sub>	
silver hyposuffite	NHGI	
gallium cyanate	NaNO	
barium iodate	Ble(CPO <sub>2</sub>	
	Name or write the formula for the following suitable intribe perphosphate introduced prophosphate percarbonate in programmed in the formula for the following permanganate carbonite. In the formula for the following benefits in the following benefits	Name or write the formula for the following polyatomic ions suiface CO <sub>2</sub> <sup>2</sup> MinO <sub>3</sub> <sup>1</sup> SO <sub>2</sub> <sup>2</sup> MinO <sub>3</sub> <sup>1</sup> SO <sub>2</sub> <sup>2</sup> SO <sub>2</sub> <sup>2</sup> SO <sub>3</sub> <sup>2</sup> SO <sub>4</sub> <sup>2</sup> SO <sub>5</sub> <sup>2</sup> SO <sub></sub>

# **Polyatomic Ions Extension Questions Answers**

**J Dewey** 

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