

Polar Bears and Penguins



Directions: Read the picture above to answer the following questions.

1. Draw the Lewis Dot Structure for HCl. Place drawing below.



2. If that penguins represents the hydrogen atom and the polar bear represents the chlorine atom, what does the dot represent in the drawing?

a. It is the bonded pair of electrons (chlorine, hydrogen)

3. What do the polar bears represent in trying to illustrate?

a. Electronegativity between atoms

Directions: Read the [Polar Bear and Penguin Analogy](#), answer the following questions as you read.

1. In polar bear analogy, define electronegativity.

a. Ability of an atom to attract bonding electrons towards itself

2. Polar bears like to eat. What do they eat in polar molecules?

a. A molecule with a difference in electrical charge

3. Fill out the table below to answer the following questions: what does hydrogen eat, chlorine eat, represent and why? Is it like a bear? What happens in that bonding, electrons are more likely to be with?

Type of Bond	What happens to the bonding electrons in each type of bond?
Nonpolar covalent	The electrons do not spend more time around either of the atoms.

Polar Bears And Penguins Chemistry Answers

New Scientist



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Provided by publisher Penguins and Polar Bears Sandra Lee Crow,2000-07-06 Describes the physical characteristics and life cycle of two animals adapted for survival in polar climates the penguin at the Antarctic and the polar bear at the Arctic
 Do Polar Bears Get Lonely? New Scientist Magazine Staff,Mick O'Hare,2009-12-22 **The Polar Bear and Penguin Will Never Meet!** Mark Oblinger,2017

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