







Types of Chemical Reactions Do atoms rearrange in predictable patterns during chemical reactions?

Why?

Recognizing patterns allows us to predict future behavior. Weather experts use patterns to predict dangerous sources so people can get their families to safety. Political analysts use patterns to predict election ourcomes. Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions.

Model 1 - Types of Reactions

$$4Fe(s) + 3O_s(g) \rightarrow 2Fe_sO_s(s)$$

$$N_2(g) + 3H_2(g) \rightarrow 2NH_1(g)$$

$$MgO(x) + H_xO(1) \rightarrow Mg(OH)_x(uq)$$

$$P_rO_r(g) = 3H_rO(l) \rightarrow 2H_rPO_r(aq)$$

$$SO_*(g) + H_*O(f) \rightarrow H_*SO_*(aq)$$

Set C ..

$$2FeCl_{s}(aq) + 3Ze(s) \rightarrow 2Fe(s) + 3ZeCl_{s}(aq)$$

$$2Al(NO_s)_s(aq) + 3Ca(s) \rightarrow 3Ca(NO_s)_s(aq) + 2Al(s)$$

$$Mg(s) + CuSO_{j}(aq) \rightarrow MgSO_{j}(aq) + Cu(s)$$

$$2AI(s) + 6HCI(aq) \rightarrow 2AICI_s(aq) + 3H_s(g)$$

$$Cl_s(g) + 2NaBe(aq) \rightarrow 2NaCl(aq) + Br_s(l)$$

$$ZnBr_{r}(aq) + F_{r}(g) \rightarrow ZnF_{r}(aq) + Br_{r}(0)$$

$$M_{H}CO_{s}(s) \rightarrow M_{H}CO(s) + CO_{s}(g)$$

$$2H_sO(b) \rightarrow 2H_s(g) + O_s(g)$$

$$2KCBO_{s}(s) \rightarrow 2KCB(s) + 3O_{s}(g)$$

$$2Na_iO_i(s) \rightarrow 2Na_iO(s) + O_i(g)$$

$$(NH_{\star})_1OO_{\star}(s) \rightarrow 2NH_{\star}(s) + H_{\star}O(t) + OO_{\star}(s)$$

Sec ID:

$$AgNO_{s}(aq) + NaCl(aq) \rightarrow AgCl(s) + NaNO_{s}(aq)$$

$$Na_sCO_s(aq) + CaCI_s(aq) \rightarrow$$

 $CaCO_s(s) + 2NaCI(aq)$

$$FeS(s) + 2HCl(aq) \rightarrow H_sS(g) + FeCl_s(aq)$$

$$HCl(aq) + NaOH(aq) \rightarrow H_1O(l) + NaCl(aq)$$

$$FeBr_s(aq) * K_sPO_s(aq) \rightarrow FePO_s(s) * 3KBr(aq)$$

The chemical equations in Model 1 contain the phase notations (s), (l), (g), and (aq). Match
each symbol with its meaning.

- 2. Based on the examples provided, which set(s) of reactions in Model 1 typically involve ions in solution (A, B, C or D)?

 SEF (and)
- 3. Based on the examples provided, which set(s) of seactions in Model 1 typically involve gases and/or solids?

 A CKA B

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