

Lab POGIL: Metal Reactivity – Introduction & Pre-Lab

Discovery Question: *How can trends in metal reactivity be used to predict the products of chemical reactions?*

Objectives:

- Develop a procedure to test the reactivity of metals
- List metals in order of reactivity based on the results of your procedure
- Predict the products of chemical reactions by considering the relative reactivity of the reactants

You have been provided with vials containing 3 types of metals (A, B, C). You also have vials filled with solutions containing cations of each metal (A^{+} , B^{+} , C^{+}), and an additional vial containing a solution of cation D^{+} . You will also have (6-well x 4-well) plates available for small-scale reactions of metals and ionic solutions.

Elemental Metals	Solutions Containing Metal Cations
A	A (contains A^{+})
B	B (contains B^{+})
C	C (contains C^{+})
	D (contains D^{+})

Remember to wear goggles at all times. Avoid skin contact with all solutions. If contact occurs, rinse with water immediately. Solution D will stain skin and clothes.

Pre-lab Requirements:

- Describe a method for predicting charges of ions formed from single atoms of an element using the periodic table.
- Use a periodic table to predict charges for ions formed from each element listed below:

Element	Ion	Element	Ion	Element	Ion
Lithium (Li)	_____	Beryllium (Be)	_____	Copper (Cu)	____/____
Sodium (Na)	_____	Magnesium (Mg)	_____	Zinc (Zn)	_____
Potassium (K)	_____	Calcium (Ca)	_____	Silver (Ag)	_____
Nitrogen (N)	_____	Oxygen (O)	_____	Fluorine (F)	_____
Phosphorous (P)	_____	Sulfur (S)	_____	Chlorine (Cl)	_____
		Selenium (Se)	_____	Bromine (Br)	_____

Pogil Chemistry Reactivity Of Metals Answers

David Kirk



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Table of Contents Pogil Chemistry Reactivity Of Metals Answers

1. Understanding the eBook Pogil Chemistry Reactivity Of Metals Answers
 - The Rise of Digital Reading Pogil Chemistry Reactivity Of Metals Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Chemistry Reactivity Of Metals Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Chemistry Reactivity Of Metals Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Chemistry Reactivity Of Metals Answers
 - Personalized Recommendations
 - Pogil Chemistry Reactivity Of Metals Answers User Reviews and Ratings
 - Pogil Chemistry Reactivity Of Metals Answers and Bestseller Lists
5. Accessing Pogil Chemistry Reactivity Of Metals Answers Free and Paid eBooks
 - Pogil Chemistry Reactivity Of Metals Answers Public Domain eBooks
 - Pogil Chemistry Reactivity Of Metals Answers eBook Subscription Services
 - Pogil Chemistry Reactivity Of Metals Answers Budget-Friendly Options
6. Navigating Pogil Chemistry Reactivity Of Metals Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Pogil Chemistry Reactivity Of Metals Answers Compatibility with Devices
 - Pogil Chemistry Reactivity Of Metals Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil Chemistry Reactivity Of Metals Answers
 - Highlighting and Note-Taking Pogil Chemistry Reactivity Of Metals Answers
 - Interactive Elements Pogil Chemistry Reactivity Of Metals Answers
 8. Staying Engaged with Pogil Chemistry Reactivity Of Metals Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pogil Chemistry Reactivity Of Metals Answers
 9. Balancing eBooks and Physical Books Pogil Chemistry Reactivity Of Metals Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Chemistry Reactivity Of Metals Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Pogil Chemistry Reactivity Of Metals Answers
 - Setting Reading Goals Pogil Chemistry Reactivity Of Metals Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Pogil Chemistry Reactivity Of Metals Answers
 - Fact-Checking eBook Content of Pogil Chemistry Reactivity Of Metals Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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