

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

1. Physical Science Basics

- What tool(s) would you want to use to find the:
 - Mass of an object? triple beam balance Basic SI Unit? kg
 - Volume of an object? ruler or water displacement method Basic SI Unit? cm³ or ml
 - Length of an object? Ruler Basic SI Unit? meter
 - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
 - What is the density of a cube that is 25 g and has a volume of 3 cm³? 8g/cm³
 - Compare the density of solid H₂O to liquid H₂O, and explain what observable effect this has.
solid H₂O less dense than liquid H₂O

2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
 - An element's properties can be predicted from its location in the periodic table.
 - The atomic number represents the number of protons.
 - The atomic mass represents the number of neutrons and protons.

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O García



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systems are crucial to public safety for years the designers of those systems had few published resources to reference and guide them through their design processes The first edition of this book changed all that and now The Design and Layout of Fire Sprinkler Systems Second Edition suits their needs even better Written and thoroughly updated by a fire prevention engineer with more than 20 years of experience this book provides a complete systematic introduction to automatic fire sprinkler design and layout from design basics code requirements and pipe hanging to hydraulic calculations retrofits and details on fire pumps The author carefully outlines all of a designer s responsibilities and includes an entire chapter dedicated to preparing for the NICET exam More than 150 sample diagrams checklists sample forms spec sheets photographs and a glossary complement the text and the larger page size of this edition permits clear presentation of diagrams and schematics The Design and Layout of Fire Sprinkler Systems not only builds the foundation and skills of newcomers to the field but also provides an outstanding reference for fire safety professionals building inspectors insurance underwriters and municipal officials **Chemical News and Journal of Physical Science** ,1895 *Physics Handbook & Study Guide Grade 11-12 IEB* Kevin Smith,2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical

summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text , **El-Hi Textbooks in Print** ,1984 *Educational Times* ,1896 **The Chemical News and Journal of Physical Science** ,1903 Computer Supported Collaborative Learning 2005 Timothy Koschmann,2017-10-03 The Computer Supported Collaborative Learning CSCL conference has become an internationally recognized forum for the exchange of research findings related to learning in the context of collaborative activity and the exploration of how such learning might be augmented through technology This text is the proceedings from CSCL 2005 held in Taipei Taiwan This conference marked the 10th anniversary of the first CSCL Conference held at Indiana University in 1995 Subsequent meetings have been held at the University of Toronto Stanford University University of Maastricht Netherlands University of Colorado at Boulder and the University of Bergen Norway Just as the first CSCL conference was instrumental in shaping the trajectory of the field in its first decade the conference in Taipei will play an important role in consolidating an increasingly international and interdisciplinary community and defining the direction of the field for the next 10 years This volume and the papers from which it is comprised will be an important resource for those active in this area of research and for others interested in fostering learning in settings of collaboration CliffsTestPrep CSET: Multiple Subjects Jerry Bobrow,Stephen Fisher,2007-05-04 The CliffsTestPrep series offers full length practice exams that simulate the real tests proven test taking strategies to increase your chances at doing well and thorough review exercises to help fill in any knowledge gaps CliffsTestPrep CSET can help you prepare for the California Subject Examination Test Multiple Subjects The Commission on Teacher Credentialing uses the CSET to evaluate subject matter competence for instructors seeking the Multiple Subject Teaching Credential Inside this test prep tool you ll find Full length practice tests with answers and in depth explanations Analysis of exam areas and question types with emphasis on suggested approaches and samples Intensive review of subjects using outlines glossaries and diagnostic tests Introduction to the format and scoring of the exam overall strategies for answering multiple choice questions and questions commonly asked about the CSET Some test taking tips and reminders to put candidates on the right track This book will help you understand the types of questions that will test your knowledge in seven general areas including Visual and Performing Arts You can get ready to show what you know in topics such as Sentence structure preferred usage and conventional forms of spelling capitalization and punctuation in written English United States and California history of early exploration through modern day economic political and cultural development The fundamentals of mathematics with focus on prime numbers factors integers ratio area volume perpendicular and more Primary scientific concepts principles and interrelationships in the context of real life problems and significant science phenomena and issues Concepts of biomechanics that affect movement and the critical elements of basic movement skills Social development of children and young adolescents including persons with special needs Components of dance music theatre and visuals arts education With guidance from the CliffsTestPrep series you ll feel

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