



# Perkin Elmer Aas 400 Manual

**D Keegan**



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2 x 35 453 297 g 1 10 g de  $\text{RaCl}_2$  tem 2 03 x 1020 moléculas de  $\text{RaCl}_2$  ou 2 03 x 1020 átomos núcleos de Ra b

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fração de carbono  $^{14}\text{C}$  isótopo radioativo do carbono que a atmosfera

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 $^{14}\text{C}$  é de 5730 anos então também sabemos a constante de decaimento e portanto o tempo total de decaimento  $t$  solução nós  
reorganizamos a equação ref lei de decaimento  $N = N_0 e^{-\lambda t}$  para dar

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de sua taxa inicial de decaimento soluÇÃo  $\tau$  140 dias  $\tau$  0 693  $\lambda = \frac{\ln 2}{\tau} = \frac{\ln 2}{0 693 \times 140} = 0 00866 \text{ dia}^{-1}$   
 $\lambda t = \ln 2$   $t = \frac{\ln 2}{\lambda} = \frac{\ln 2}{0 00866} = 80 1 \text{ dias}$

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vim 3800 anos ou temos exemplo do radônio  $^{220}\text{Rn}$  vim 79s para o radônio  $^{220}\text{Rn}$  a cada 79 segundos um átomo deve se  
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un exceso de partículas emite partículas y o radiación hasta alcanzar una configuración estable el exceso de partículas se  
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partícula alfa emissão de partícula beta emissão de pósitron captura de elétron e fissão espontânea cada um desses tipos de

decaimento está relacionado com as características da estrutura nuclear de um dado radionuclídeo

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