# Periodic Trends

Directions: Choose the letter that corresponds to the correct answer.

- Which of the following can be considered as Dobereiner's triad?
  - a. He: 4.0. Lt: 6.9. Na: 23.0
  - b. N: 14.0, P: 40.0, At 217.0
  - Ne: 20.2, Ar. 39.9, Kr. 83.8
  - d. Ca: 40.1, Sr. 87.6, Ba: 137.3
- 2. Which of the following elements does not belong to the group?
  - a. N
  - b S
  - c. Po
  - d. Ter
- 3. Which of the following is/are TRUE about boron and oxygen?
  - a. both are metalloids
  - b. boron is smaller than oxygen.
  - boron is more electronegative than oxygen
  - d. higher energy is required to eject an electron from the valence shell of boron.
  - e. all of the above
  - f. none of the above

For 4 and 5, consider Na, Mg Al, and Si.

- The element that ejects one electron from its valence shell most readily is
  - a Na
  - b. Ma
  - C. All
  - d. Si
- 5. Which element has the most negative electron affinity?
  - e. Na
  - f. Mg



# **Periodic Trends Practice Questions**

**The Princeton Review** 

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