

1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.*

2. Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.*

3. Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu **K** Ni **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.*

4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

- 5.
- O C **Be** **Ne** Same Period
 - Na Rb **Fr** **H** Same Group
 - Pb** **C** Sn Si Same Group
 - Au W S **Fr** **Ne** Zn Challenge

6. For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

- O C **Be** **Ne** Same Period
- Na Rb **Fr** **H** Same Group
- Pb** **C** Sn Si Same Group
- Au W S **Fr** **Ne** Zn Challenge

Periodic Trends Practice Answers

Morris Hein, Susan Arena, Cary Willard



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Periodic Trends Practice Answers :

Exemplars Exemplar 1: Topic 8: An analysis and evaluation of the business and financial performance of an organisation over a three year period. Exemplars Many of the key themes from the ACCA syllabus - particularly financial reporting, performance measurement and business analysis - have been discussed in this ... OXFORD BROOKES BUSINESS SCHOOL - cloudfront.net Feb 19, 2018 — Business School, Oxford Brookes University. MESSAGE FROM THE VICE-CHANCELLOR. Oxford Brookes University and by extension Oxford. Brookes ... THE FACULTY OF BUSINESS - cloudfront.net with recent examples on green reporting, business ethics, stakeholder ... OXFORD BROOKES UNIVERSITY FACULTY OF BUSINESS. 10.

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