

The Periodic Table

The Structure of the Periodic Table

- The periodic table displays the elements in order of increasing atomic number i.e. in order of increasing number of protons.
- Elements are arranged in 18 vertical columns called groups that are numbered 1 – 18.
- The seven horizontal rows in periodic table are called periods. The periods are numbered 1-7. The first period contains only two elements, hydrogen and helium.
- Elements within the same groups have similar chemical properties because they have the same number of valence electrons. The number of valence electrons is the same as the group number (in Roman numerals).

		GROUPS 1 - 18																					
PERIODS 1 - 7	1	H 1.0079																	He 4.0026				
	2	Li 6.941	Be 9.0122															B 10.811	C 12.011	N 14.007	O 15.999	F 18.998	Ne 20.18
	3	Na 22.99	Mg 24.305															Al 26.982	Si 28.086	P 30.974	S 32.06	Cl 35.453	Ar 39.948
	4	K 39.098	Ca 40.078	Sc 44.956	Ti 47.88	V 50.942	Cr 51.996	Mn 54.938	Fe 55.847	Co 58.933	Ni 58.693	Cu 63.546	Zn 65.39	Ga 69.723	Ge 72.64	As 74.922	Se 78.96	Br 79.904	Kr 83.8				
	5	Rb 85.468	Sr 87.62	Y 88.906	Zr 91.224	Nb 92.906	Mo 95.94	Tc 97.907	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.6	I 126.9	Xe 131.29				
	6	Cs 132.91	Ba 137.33	La 138.91	Hf 178.49	Ta 180.95	W 183.84	Re 186.21	Os 190.23	Ir 192.22	Pt 195.08	Au 196.97	Hg 200.59	Tl 204.38	Pb 207.2	Bi 208.98	Po 209	At 210	Rn 222				
	7	Fr 223	Ra 226	Ac 227	Rf 261	Ha 262	Sg 266	Nh 271	Hs 277	Mt 288	Ds 289	Uus 294	Uub 295										

Energy levels and period numbers

The electrons of an atom are arranged in energy levels. The period number tells us in which energy level an atom has its valence electrons.

For example:

Element	In which Group?	In which Period?	How many valence electrons	Which energy level are valence electrons?
Phosphorous	15	3	5	3 rd energy level
Potassium				
Calcium				
Oxygen				

Periodic Table Lessons Fourth Grade

S Ben Porath



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Lesson Design for Differentiated Instruction, Grades 4-9 Kathy Tuchman Glass, 2009-01-14 This user friendly resource provides step by step guidance and a detailed template for creating meaningful lessons that are differentiated according to students learning characteristics [Arun Deep's Self-Help to ISC Chemistry Class 11 : For 2025-26 Examinations](#) Amar Nath Bhutani, Saurabh Joshi, 2025-07-07 Arun Deep's Self Help to ISC Chemistry Class 11 For 2025 26 Examinations This guidebook has been meticulously crafted to support students of Class 11 who are preparing for the ISC Chemistry examination for the academic year 2025 26 Aligned with the latest ISC curriculum the book provides comprehensive solutions and explanations to all the questions presented in the ISC Chemistry textbook published by Nageen Prakashan The content is structured to aid conceptual clarity reinforce theoretical understanding and strengthen problem solving skills Each chapter includes Detailed answers to all in text and end of chapter questions Step by step solutions for numerical problems Additional tips and key points for effective revision Supportive content that complements classroom learning An ideal companion for ISC students this Self Help book aims to simplify complex concepts and provide exam oriented preparation helping learners achieve academic excellence with confidence **ISC Chemistry Book 1 for Class XI (2021 Edition)** R.D.MADAN, ISC Chemistry Book 1 *All in 1 Guide Book: CBSE Class 10 for 2022 Examination* Gurukul, 2021-06-21 Benefit from Effective Practice Easy Revision for Class 10 CBSE Board Examinations 2022 with our All in 1 Guide Book Consisting of 6 subjects including English Language Literature Hindi A Hindi B Mathematics Science and Social Science Our handbook will help you study well and prepare at home with all the answers strictly based on marking scheme issued by Board Why should you trust Gurukul Books Oswal Publishers Oswal Publishers has been in operation since 1985 Over the past 30 years we have developed content that aids students and teachers in achieving excellence in education We create content that is extensively researched meticulously articulated and comprehensively edited catering to the various National and Regional Academic Boards in India How can you benefit from Gurukul All in 1 Guide Book for 10th Class Our handbook is a one stop solution for Class 10 CBSE students study requirements With multiple subjects in one book formulated chapterwise and categorywise also including NCERT Past Years Board Examination Papers Toppers's Answers our guide is a complete book you will need to prepare for 2022 board examinations Apart from study material and solved papers in 6 subjects our book is enriched with MCQs Probable Objective Type Questions to improve study techniques for any exam paper Students can create vision boards to establish study schedules and maintain study logs to measure their progress With the help of our handbook students can also identify patterns in question types and structures allowing them to cultivate more efficient answering methods Our book can also help in providing a comprehensive overview of important topics in each subject with Source based Case based Passage based and Picture based Questions making it easier for students to prepare for the exams **Conceptual Chemistry Class XI Vol. I** S K Jain, A book on Conceptual Chemistry *Self-Help to I.C.S.E.*

Chemistry Class 10 (For 2022-23 Examinations) Amar Bhutani, RP Gupta K., This book is written strictly in accordance with the latest syllabus prescribed by the Council for the I C S E Examinations in and after 2023 This book includes the Answers to the Questions given in the Textbook Concise Chemistry Class 10 published by Selina Publications Pvt Ltd This book is written by Sunil Manchanda

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