

5.1 Atomic structure and the Periodic Table

You will learn:

- To understand how the position of an element in the Periodic Table can be used to predict its atomic structure and chemical and physical properties.
- To understand that models and analogies can change according to scientific evidence
- To describe the uses, strengths and limitations of models and analogies
- To represent scientific ideas using recognised symbols or formulae

1.

Draw a line to match the word with its correct meaning.

The first line has been drawn for you.

Word	Meaning
Electrons	Particles with no charge found in an atom
Neutrons	The central part of the atom
Protons	The paths or orbits that electrons move along in the atom
Nucleus	Very small negatively charged particles in an atom
Shells	Particles with a positive charge found in an atom

2.

Complete the sentences.

Use the words in the box.

atomic	mass	protons	electrons	nucleus
	shells	neutrons	atom	

- a An _____ number is the protons an _____ of an element has.
- b A _____ number is the total number of _____ and _____ found in the nucleus of an atom.

Periodic Table Workbook Answers Chemistry

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Svarney**



Periodic Table Workbook Answers Chemistry:

The Enigmatic Realm of **Periodic Table Workbook Answers Chemistry**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Periodic Table Workbook Answers Chemistry** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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