

2.6 - Periodic Trends worksheet

Name _____

I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

II. Answer the questions using the symbols on the periodic table

1. What elements are in period 5?

What element(s) is a

- | | |
|-------------------------------|---------------------------|
| _____ 2.halogens | _____ 6.transition metals |
| _____ 3. lanthanides | _____ 7.alkali metals |
| _____ 4.alkaline earth metals | _____ 8.actinides |
| _____ 5.noble gases | |

What elements are in the sublevel

- _____ 9. s _____ 11. d
_____ 10. p _____ 12. f

What elements are

- _____ 13. metals,
- _____ 14. nonmetals
- _____ 15. metalloids,
- _____ 16. Which element is the most reactive nonmetal?
- _____ 17. Which element is the most reactive metal?
- _____ 18. Which element below is the least reactive?
- _____ 19. Which element has the largest radius in period 4?
- _____ 20. Which element has the largest radius in the Alkali metals family?
- _____ 21. Which element has the highest ionization energy in period four?
- _____ 22. Which elements has the second largest atomic number?
- _____ 23. Which element has the largest atomic mass?
- _____ 24. Which element is located at period 4, group 15?
- _____ 25. Which element is located at period 5, group 2?
- _____ 27. Which elements will form +1 ions?
- _____ 28. Which group of metals will have the same properties
- _____ 29. Which group of nonmetals will have the same properties
- _____ 30. On the back, define all chapter 5, section 3 vocabulary terms.

[illegible][illegible]

Periodic Table Chemistry Answers

Christian G. Meyer



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Chemistry Questions - Answers for Intermediate Mr. Ajeet Singh, 2018-05-13 A question answer formatted book designed for intermediate level chemistry students covering theory based and numerical questions to aid learning and exam preparation

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