

## 7-3 Review and Reinforcement

### Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

| Number | Prefix |
|--------|--------|
| 1      | mono   |
| 2      | di     |
| 3      | tri    |
| 4      | tetra  |
| 5      | penta  |
| 6      | hexa   |
| 7      | hepta  |
| 8      | octa   |
| 9      | nona   |
| 10     | deca   |

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation  
anion

hydrates  
anhydrates

electronegative  
hydrogen ions

ionic  
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate—**.
- When naming an ionic substance, the name of the **—cation—** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **—electronegative—** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **—hydrogen ions—**.
- The name of an acid is usually derived from the name of its characteristic **—anion—**.
- Prefixes are often used when naming **molecular** compounds.

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**KJ Lindholm-Leary**



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