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F.Y.B.Sc. (Biotechnology)

CHEMISTRY

Bb - 101 : Fundamentals of Chemistry
(2008 Pattern)

Time : 3 Hours]

[Max. Marks : 80

Instructions to the candidates:

- 1) All questions are compulsory.
- 2) Draw neat and labelled diagrams wherever necessary.
- 3) Figures to the right indicates full marks.
- 4) Use of logarithmic table and calculator is allowed.

Q1) Answer the following :

[16]

- a) A sample of Helium has volume of 520 mL at 100°C. Calculate the temperature at which the volume will become 260 mL. Assume that pressure remains constant.
- b) Distinguish between order and molecularity.
- c) What is meant by abnormal molecular weight.
- d) Define the term component. How many components are present in the following system :



- e) What is effect of dilution of equivalent conductance of strong electrolyte.
- f) How will you determine the solubility product of sparingly soluble salt by emf method?
- g) What are distereomers?
- h) Distinguish between sigma (σ) and pi (Π) bond.

Q2) Attempt any FOUR of the following :

[16]

- a) State and explain Dalton's law of partial pressure.
- b) Show that half life period of first order reaction is independent of initial concentration of reactants.
- c) What are colligative properties? Explain any one colligative property.
- d) Describe KI - H₂O system on the basis of phase rule.
- e) Discuss Debye - Huckel theory of activity coefficients of strong electrolyte.
- f) Explain the factors affecting Vander Waal's forces.

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1. Understanding the eBook Nirali For Fybsc Chemistry 2014
 - The Rise of Digital Reading Nirali For Fybsc Chemistry 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Nirali For Fybsc Chemistry 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nirali For Fybsc Chemistry 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nirali For Fybsc Chemistry 2014
 - Personalized Recommendations
 - Nirali For Fybsc Chemistry 2014 User Reviews and Ratings
 - Nirali For Fybsc Chemistry 2014 and Bestseller Lists
5. Accessing Nirali For Fybsc Chemistry 2014 Free and Paid eBooks
 - Nirali For Fybsc Chemistry 2014 Public Domain eBooks
 - Nirali For Fybsc Chemistry 2014 eBook Subscription Services
 - Nirali For Fybsc Chemistry 2014 Budget-Friendly Options
6. Navigating Nirali For Fybsc Chemistry 2014 eBook Formats

- ePub, PDF, MOBI, and More
- Nirali For Fybsc Chemistry 2014 Compatibility with Devices
- Nirali For Fybsc Chemistry 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nirali For Fybsc Chemistry 2014
 - Highlighting and Note-Taking Nirali For Fybsc Chemistry 2014
 - Interactive Elements Nirali For Fybsc Chemistry 2014
- 8. Staying Engaged with Nirali For Fybsc Chemistry 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nirali For Fybsc Chemistry 2014
- 9. Balancing eBooks and Physical Books Nirali For Fybsc Chemistry 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nirali For Fybsc Chemistry 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nirali For Fybsc Chemistry 2014
 - Setting Reading Goals Nirali For Fybsc Chemistry 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nirali For Fybsc Chemistry 2014
 - Fact-Checking eBook Content of Nirali For Fybsc Chemistry 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
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