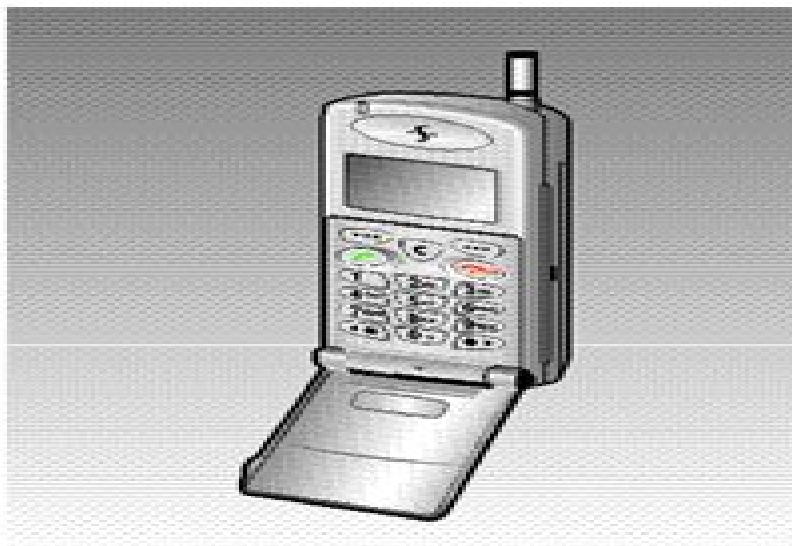


**SAMSUNG**

**DUAL BAND Mobile  
Cellular Phone  
SGH-2488**

# ***SERVICE* Manual**

**DUAL BAND Mobile Cellular Phone**



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# Samsung Sgh 2488 Service Manual

**Zenith Radio Corporation**



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web the  $\text{A}^-$  ion is a base because it is accepting a proton from the  $\text{H}_2\text{O}$  the reversible reaction contains two acids and two bases we think of them in pairs called conjugate pairs when the acid  $\text{HA}$  loses a proton it forms a base  $\text{A}^-$  when the base  $\text{A}^-$

accepts a proton back again it obviously reforms the acid

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