

Chemistry pH and pOH Worksheet

1. Calculate the values of both pH and pOH of the following solutions.

	pH	pOH
0.020 M HCl		
0.0050 M NaOH		
A blood sample 7.2×10^{-8} M of H^+		
0.0050 M NaOH		

2. Find the values of $[H^+]$, pOH, and $[OH^-]$ that correspond to each of the following pH values.

	$[H^+]$	$[OH^-]$	pOH
pH of lemon juice = 2.9			
pH of sauerkraut = 3.85			
pH of milk of magnesia = 10.81			
pH of orange juice = 4.11			
pH of diluted household ammonia = 11.61			

3. Determine which of the solutions in problem #2 are acidic.
4. A certain brand of root beer has a hydrogen concentration equal to 1.9×10^{-5} M. What are the pH and pOH of this root beer?
5. Dr. Pepper has a $[H^+] = 1.4 \times 10^{-5}$ M. What is its pH?

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Table of Contents Poh Practice Problems

1. Understanding the eBook Poh Practice Problems
 - The Rise of Digital Reading Poh Practice Problems
 - Advantages of eBooks Over Traditional Books
2. Identifying Poh Practice Problems
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Poh Practice Problems
 - User-Friendly Interface
4. Exploring eBook Recommendations from Poh Practice Problems
 - Personalized Recommendations
 - Poh Practice Problems User Reviews and Ratings
 - Poh Practice Problems and Bestseller Lists
5. Accessing Poh Practice Problems Free and Paid eBooks

- Poh Practice Problems Public Domain eBooks
- Poh Practice Problems eBook Subscription Services
- Poh Practice Problems Budget-Friendly Options
- 6. Navigating Poh Practice Problems eBook Formats
 - ePub, PDF, MOBI, and More
 - Poh Practice Problems Compatibility with Devices
 - Poh Practice Problems Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Poh Practice Problems
 - Highlighting and Note-Taking Poh Practice Problems
 - Interactive Elements Poh Practice Problems
- 8. Staying Engaged with Poh Practice Problems
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Poh Practice Problems
- 9. Balancing eBooks and Physical Books Poh Practice Problems
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Poh Practice Problems
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Poh Practice Problems
 - Setting Reading Goals Poh Practice Problems
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Poh Practice Problems
 - Fact-Checking eBook Content of Poh Practice Problems
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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