

## POGIL: Lewis Dot Structures Part 1 (20 P)

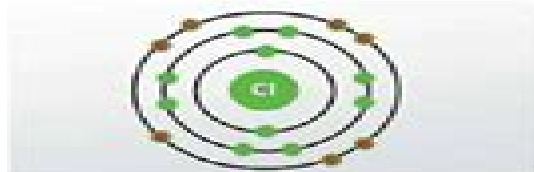
How are electrons shared to create covalently bonded molecules?

### Read This!

**Why?** Covalent bonds result from a *sharing* of electrons by two or more atoms (usually nonmetals). **Lewis theory** (Gilbert Newton Lewis, 1875-1946) focuses on the valence electrons, since the outermost electrons are the ones that are highest in energy and farthest from the nucleus, and are therefore the ones that are most exposed to other atoms when bonds form. **Lewis dot diagrams** for elements are a handy way of picturing valence electrons, and especially, which electrons are available to be shared in covalent bonds.

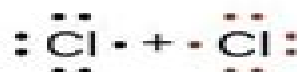
#### Bohr Model: Chlorine

7 valence electrons on outermost energy level



#### Lewis Dot diagram: Chlorine

7 valence electrons



chlorine  
atoms



chlorine  
molecule



#### Lewis Dot Diagrams Showing Valence Electrons for Individual Elements

| 1                | 2                     |                   | 3                     | 4                     | 5                    | 6                    | 7                     | 8                     |
|------------------|-----------------------|-------------------|-----------------------|-----------------------|----------------------|----------------------|-----------------------|-----------------------|
| $\cdot\text{H}$  |                       |                   |                       |                       |                      |                      |                       | $\text{He}$           |
| $\cdot\text{Li}$ | $\cdot\text{Be}\cdot$ |                   | $\cdot\text{B}\cdot$  | $\cdot\text{C}\cdot$  | $\cdot\text{N}\cdot$ | $\cdot\text{O}\cdot$ | $\cdot\text{F}\cdot$  | $\cdot\text{Ne}\cdot$ |
| $\cdot\text{Na}$ | $\cdot\text{Mg}\cdot$ | Transition metals | $\cdot\text{Al}\cdot$ | $\cdot\text{Si}\cdot$ | $\cdot\text{P}\cdot$ | $\cdot\text{S}\cdot$ | $\cdot\text{Cl}\cdot$ | $\cdot\text{Ar}\cdot$ |

Atoms are grouped together by number of valence electrons; all atoms in Group 1 have 1 valence electron.

# Pogil Answers Chemistry Lewis Dot

**Arturo Cuomo**



## **Pogil Answers Chemistry Lewis Dot:**

## Reviewing **Pogil Answers Chemistry Lewis Dot**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Pogil Answers Chemistry Lewis Dot**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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