

Nuclear Fission and Fusion

Why?

Fission and fusion reactions generate heat which can be used to generate electricity. Nuclear fission provides the energy for nuclear power plants and fusion is the source of the sun's energy. The use of fusion in power plants can help conserve fossil fuels. However, the major problem is the creation of isotopes of hydrogen, the fuel for fusion. Isotopes of hydrogen are radioactive and must be stored safely for long periods.

Learning Objectives

- 1. Explain and compare nuclear fission and fusion reactions.
- 2. Understand the conservation laws that apply to nuclear reactions.

Learning Objectives

- 1. Explain the difference between fission and fusion reactions.
- 2. Identify the major conservation laws that apply to nuclear reactions.

Terminology

- 1. Fission
- 2. Fusion
- 3. Radioactive
- 4. Isotope

Terminology

- 1. Nuclear fission and fusion.
- 2. Conservation of mass, energy and momentum.
- 3. Nuclear reactions involving fission, fusion, and decay.

Terminology

Fission is the transformation of the nucleus of an atom in which the mass is changed from the nucleus into a different nucleus. This can be accomplished through many types of reactions, including fission and fusion.

Fusion is the transformation of two nuclei into a different nucleus. Fusion can be accomplished by fission and fusion. Fusion can be accomplished by fission and fusion.

Energy and mass are conserved, but not the total energy-momentum. Energy-momentum is conserved, but not the total energy-momentum. Energy-momentum is conserved, but not the total energy-momentum.

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J Ma



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