



Republic of the Philippines
Department of Education
REGION III

REGIONAL DIAGNOSTIC ASSESSMENT IN PHYSICAL SCIENCE

Name _____ Score _____

Direction: Choose the correct answer for each item.

1. What type of nuclear synthesis is believed to be the origin of the production of heavy elements?
A. Big Bang Nucleosynthesis
B. Primordial Nucleosynthesis
C. Stellar Nucleosynthesis
D. Supernova Nucleosynthesis
2. What is the ability of an atom in a compound to attract electrons towards itself called?
A. Electromotive force
B. Electron Affinity
C. Electronegativity
D. Ionization energy
3. Which of the following is the measure of a liquid's resistance to flow?
A. Boiling point
B. Melting point
C. Solubility
D. Viscosity
4. Which of the following is the building block of protein?
A. Amino acid
B. Glycogen
C. Monomer
D. Nucleotide
5. What do you call the substance that increases the rate of chemical reaction without itself being consumed by the reaction?
A. Activation energy
B. Catalyst
C. Product
D. Reactant
6. Which of the following describes a stellar nucleosynthesis?
A. It is the process by which elements are formed within stars.
B. It is the formation of elements during a supernova explosion
C. It is the formation of light elements such as helium and hydrogen
D. It is the process by which elements are produced within gas clouds
7. Which of the following statements is TRUE about non polar molecules?
A. They have high boiling point
B. They have high melting point
C. They have low surface tension
D. They have low vapor pressure
8. Which will result in an ion- dipole interaction?
A. Attraction between an ion and a polar molecule
B. Attraction between a polar with a polar molecule
C. Repulsion between dipole and another dipole
D. Repulsion between a polar and a nonpolar molecule
9. Why are lipids insoluble in water?
A. Because Lipids are hydrophilic
B. Because Lipids are hydrophobic
C. Because Lipids are neutral
D. Because Lipids are ionic compound
10. What will be the effect of temperature rise in a chemical reaction?
A. It increases molecular concentration and slows down reaction rate
B. It increases the number of effective collisions among reactants, thus increasing the reaction rate
C. It decreases the system's energy and limits how much further reaction can take place
D. It decreases the number of collisions and the energy of molecules to limit further reaction

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Table of Contents Physical Science Department Papernovember 2014grade

1. Understanding the eBook Physical Science Department Papernovember 2014grade
 - The Rise of Digital Reading Physical Science Department Papernovember 2014grade
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Department Papernovember 2014grade
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Physical Science Department Papernovember 2014grade
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Department Papernovember 2014grade
 - Personalized Recommendations

- Physical Science Department Paper November 2014 grade User Reviews and Ratings
- Physical Science Department Paper November 2014 grade and Bestseller Lists
- 5. Accessing Physical Science Department Paper November 2014 grade Free and Paid eBooks
 - Physical Science Department Paper November 2014 grade Public Domain eBooks
 - Physical Science Department Paper November 2014 grade eBook Subscription Services
 - Physical Science Department Paper November 2014 grade Budget-Friendly Options
- 6. Navigating Physical Science Department Paper November 2014 grade eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Department Paper November 2014 grade Compatibility with Devices
 - Physical Science Department Paper November 2014 grade Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Department Paper November 2014 grade
 - Highlighting and Note-Taking Physical Science Department Paper November 2014 grade
 - Interactive Elements Physical Science Department Paper November 2014 grade
- 8. Staying Engaged with Physical Science Department Paper November 2014 grade
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Department Paper November 2014 grade
- 9. Balancing eBooks and Physical Books Physical Science Department Paper November 2014 grade
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Department Paper November 2014 grade
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Department Paper November 2014 grade
 - Setting Reading Goals Physical Science Department Paper November 2014 grade
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Department Paper November 2014 grade
 - Fact-Checking eBook Content of Physical Science Department Paper November 2014 grade

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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