

INSTRUCTIONS

This Learning Packet has two parts: (1) text to read and (2) questions to answer.

The text describes a particular sport or physical activity, and relates its history, rules, playing techniques, scoring, notes and more.

The Response Forms (questions and puzzles) check your understanding and appreciation of the sport or physical activity.

INTRODUCTION

Softball is a very popular sport in North America. Millions of children and adults play and enjoy this game. Softball has been played in the United States and Canada since the late 1800's. An International Softball Federation was formed in 1952. This organization coordinates international competition and regular regional and world championship tournaments for men and women. In 1996 softball was added as an Olympic sport. In that same year the United States became the first gold medal winner of this new Olympic sport.



The fundamentals of softball are the same as those of baseball. Bunting and fielding is similar in both games. Softball is played on a smaller field, however, and the game is seven innings long instead of nine. In softball, the ball is pitched underhand, whereas in baseball the pitch is overhand or sidearm. Base stealing is permitted in both games, but in softball the runner must keep in contact with the base until the pitcher throws a pitch.

There are many variations to the game of softball; some are officially recognized and others are not. In Chicago, for example, Windy City or Cabbage Ball is played with a 16-inch ball and without gloves. In Maine and Alaska, people play softball in the snow. In California, a game has developed called Over the Line: OTL players don't use gloves and they don't run bases. These variations of the game exist in many regions, however, fast-pitch and slow-pitch softball are the two dominate the games in North America.

Physical Education Learning Packets Softball Answer Key

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in

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