

Answer all questions on the paper/answer book provided.
A scientific calculator may be used.

QUESTION 1

There are four possible options for each answer in the following questions. Each question has only ONE correct answer. Choose the correct answer and write only A, B, C or D next to the question number.

- 1.1 A compound consists of ions X^{2+} and Y^{3-} . A possible formula for the compound could be:
- A $3X2Y$
 - B X_2Y_3
 - C X_3Y_2
 - D $(XY)_6$
- (2)
- 1.2 Which of the following is not a characteristic of a physical change:
- A no new products are formed
 - B changes can be reversed
 - C observable with the naked eye
 - D the parts of the substance lose their chemical identity
- (2)
- 1.3 Ionic bonding occurs between:
- A two non-metals
 - B two metals
 - C a metal and a non-metal
 - D a metal and a semi-metal
- (2)
- 1.4 Khumalo heats chocolate in a double boiler and observes the changes. The most accurate conclusion to his experiment is:
- A Chocolate undergoes a chemical change when it is heated
 - B Chocolate undergoes a physical change when it is heated
 - C Chocolate decomposes into its constituent components when heated
 - D Chocolate gives off latent energy when it melts
- (2)

Physical Science Control Test Term 3

Robert R. Redfield



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