

Periodic Trends Practice Worksheet

Use the periodic table and your knowledge of the periodic trends to answer the following questions.

- ① Identify each element as a metal, metalloid, or nonmetal
 - Ⓐ germanium _____
 - Ⓑ zinc _____
 - Ⓒ phosphorous _____
 - Ⓓ lithium _____
- ② Which of the two species is larger in each of the following pairs?
 - Ⓐ N^{3-} or F^- _____
 - Ⓑ Mg^{2+} or Ca^{2+} _____
 - Ⓒ Fe^{2+} or Fe^{3+} _____
- ③ Which of the two species is smaller in each of the following pairs?
 - Ⓐ K^+ or Li^+ _____
 - Ⓑ Au^+ or Au^{3+} _____
 - Ⓒ P^{3-} or N^{3-} _____
- ④ What atom has the LARGEST atomic radius?

- ⑤ What atom has the SMALLEST atomic radius?

- ⑥ Which is larger?
 - Ⓐ Mg or Mg^{2+} _____
 - Ⓑ I or I^- _____
 - Ⓒ Cl^- or Br^- _____
 - Ⓓ F^- or Ne _____
 - Ⓔ N^{3-} or P^{3-} _____
- ⑦ Put the following in order of increasing radius: Br^- Se^{2-} Sr^{2+} Kr Rb^+
- ⑧ Circle the correct answer.
 - Ⓐ Bromine has a higher / lower ionization energy than Lead
 - Ⓑ Barium has a higher / lower ionization energy than Radon
 - Ⓒ Fluorine has a higher / lower ionization energy than Arsenic
 - Ⓓ Lithium has a higher / lower ionization energy than Cesium
- ⑨ Which of the two species is larger in each of the following pairs?
 - Ⓐ Mg or Sr
 - Ⓑ Be or N
 - Ⓒ P or Cl
 - Ⓓ Ni or Br
- ⑩ Arrange the following in order of increasing electronegativity: S , O , Ne , Al

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Table of Contents Periodic Trends Practice Problems With Answers

1. Understanding the eBook Periodic Trends Practice Problems With Answers
 - The Rise of Digital Reading Periodic Trends Practice Problems With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Practice Problems With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Practice Problems With Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Practice Problems With Answers
 - Personalized Recommendations
 - Periodic Trends Practice Problems With Answers User Reviews and Ratings
 - Periodic Trends Practice Problems With Answers and Bestseller Lists
5. Accessing Periodic Trends Practice Problems With Answers Free and Paid eBooks
 - Periodic Trends Practice Problems With Answers Public Domain eBooks
 - Periodic Trends Practice Problems With Answers eBook Subscription Services
 - Periodic Trends Practice Problems With Answers Budget-Friendly Options
6. Navigating Periodic Trends Practice Problems With Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Periodic Trends Practice Problems With Answers Compatibility with Devices
 - Periodic Trends Practice Problems With Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Practice Problems With Answers
 - Highlighting and Note-Taking Periodic Trends Practice Problems With Answers
 - Interactive Elements Periodic Trends Practice Problems With Answers
 8. Staying Engaged with Periodic Trends Practice Problems With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Practice Problems With Answers
 9. Balancing eBooks and Physical Books Periodic Trends Practice Problems With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Practice Problems With Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Periodic Trends Practice Problems With Answers
 - Setting Reading Goals Periodic Trends Practice Problems With Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Periodic Trends Practice Problems With Answers
 - Fact-Checking eBook Content of Periodic Trends Practice Problems With Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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