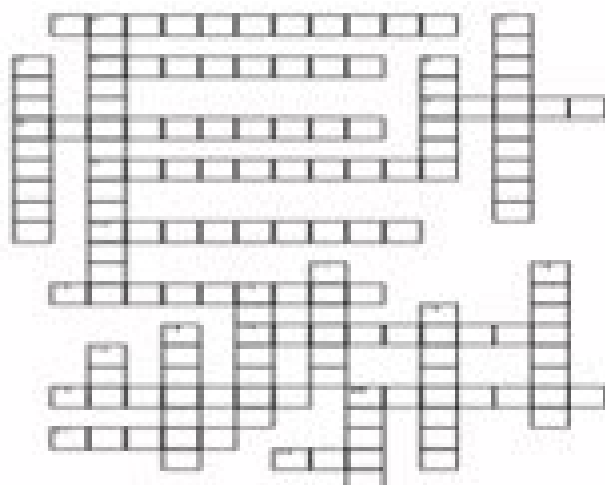


Periodic Table Assignment

Read pages 14 – 19. Complete the crossword puzzle (in pencil) and fill in the periodic table as instructed below.



- ACROSS**
1. Synthetic elements with atomic numbers greater than 92
 5. This law states that the properties of elements repeat as a function of their atomic number
 7. Silicon is in this period
 8. Elements were originally placed in order of atomic ____, now they are in order of atomic ____
 9. The group that consists of very unreactive gases
 10. Number of elements known in 1800
 13. Element 101 was named in his honour
 17. The periodic table is normally drawn with 18 columns. A more accurate representation has ____ columns
 19. Most nonmetals are in this state
 20. Nonmetal that is a liquid at SATP
 21. The general name given to a column in the periodic table
 22. According to figure 6, when aluminum combines with oxygen, three oxygen atoms combine with this many aluminum atoms

- DOWN**
2. Elements that most closely follow the periodic law
 3. The first scientist to organize elements based on repeating patterns
 4. The modern name for aka-silicon
 6. Most elements are ____, since they fall to the left of the "staircase line"
 11. The general name given to a row (left-to-right) in the periodic table
 12. Iodine belongs to this group
 14. The first alkali metal
 15. This element was assigned a relative atomic mass of one because it is the lightest element
 16. The state of matter represented by (aq)
 18. Silicon is in this group
 20. The staircase that separates metals and nonmetals begins below this element

1	H	He																	Fr	Ra
2	Li	Be	B	C	N	O	F	Ne									Rn			
3	Na	Mg	Al	Si	P	S	Cl	Ar									Ra			
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rn	
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Rn	
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
7	Fr	Ra	Ac	Th	Pa	U	Np	Pu	A	B	C	D	E	F	G	H	I	J	K	L

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tm	Yb	Lu
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es

- 1) Place a large G in boxes that house elements that are gases at SATP, and a large L in boxes that contain elements that are liquids. The rest of the elements are solids; do not write anything in these boxes.
- 2) Darken the "staircase line" that divides metals from non-metals
- 3) Metalloids (i.e. semi-conductors) are elements that have properties midway between metals and non-metals. The metalloids are: B, Si, Ge, As, Sb, Te, Po, At. Identify the metalloids by drawing diagonal lines in these boxes.
- 4) Notice that each group (column) in the periodic table is identified at the top with a number and/or letter. There are two conventions used in the textbook. Label the groups that end in A (e.g. 1A, 11A, 11A, etc).
- 5) Frame and color the following groups (note: H is not an 1A metal): Alkali metals, alkaline earth metals, halogens, noble gases, transition metals, the inner transition elements (made up of lanthanides and actinides).
- 6) Create a legend for your periodic table.

Periodic Table Assignment Answer Key

JE Gale



Periodic Table Assignment Answer Key:

Assessing Middle and High School Mathematics & Science Sheryn Spencer-Waterman,2013-08-16 For middle and high school teachers of mathematics and science this book is filled with examples of instructional strategies that address students readiness levels interests and learning preferences It shows teachers how to formatively assess their students by addressing differentiated learning targets Included are detailed examples of differentiated formative assessment schedules plus tips on how to collaborate with others to improve assessment processes Teachers will learn how to adjust instruction for the whole class for small groups and for individuals They will also uncover step by step procedures for creating their own lessons infused with opportunities to formatively assess students who participate in differentiated learning activities

College Chemistry I Norman H. Nachtrieb,1960 *Learning Chemistry 7 Solution Book (Year 2023-24)* ,2024-01-02

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Eye Tracking for STEM Education Research: New Perspectives Pascal Klein, Martin Rusek, Maike Schindler, 2024-04-04 A modern approach to improving education uses the components of experimental scientific research practices based on objective data dissemination of results and the use of modern technologies STEM education research is maturing and new tools and analysis techniques become available As one example eye tracking the recording of persons eye movements has been growing in popularity as it enables researchers to study learning materials effectiveness problem solving and even students approaches during experimentation Eye movements as captured using eye tracking can reveal information about a student s attention and cognition on a process level going well beyond classical product based assessment techniques such as questionnaires or tests *College Chemistry*

II Norman H. Nachtrieb, 1960 [Inorganic Chemistry-II \(For M.Sc. Course for Universities in Uttarakhand\)](#) Dr. Geeta Tiwari, This book entitled Inorganic Chemistry II is an effort to present the subject matter in a comprehensible and easily understandable form This textbook is purposefully prepared for the postgraduate Inorganic Chemistry second semester course and it covers all the topics recommended *Science Scope*, 2003 **THE JOURNAL**, 1999 **Chemistry in the Laboratory** James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations **Study Guide** Woolfolk, 2000-06

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