

- ☐ As more protons are added, the more tightly the surrounding electrons are drawn inwards.
- ☐ As more neutrons are added, the more tightly the surrounding electrons are drawn inwards.
- ☐ As more electrons are added, the more tightly they draw themselves towards the nucleus.
- ☐ As the mass of the nucleus rises, the more tightly the surrounding electrons are drawn inwards.

- ☐ It decreases.
- ☐ It decreases, then increases.
- ☐ It remains constant.
- ☐ It increases, then decreases.
- ☐ It increases.

The periodic table displays elements from Hydrogen (H) to Oganesson (Og). Each element is represented by its symbol, atomic number, and name. The table is organized into rows (periods) and columns (groups). The elements are color-coded by their chemical properties: alkali metals (blue), alkaline earth metals (orange), transition metals (green), post-transition metals (yellow), metalloids (purple), nonmetals (pink), and noble gases (light blue).

- ☐ Argon (Ar)
- ☐ Phosphorus (P)
- ☐ Sodium (Na)
- ☐ Aluminum (Al)
- ☐ Magnesium (Mg)

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# Periodic Elements Need To Knows Answer Key

**Camilla Rothe**



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