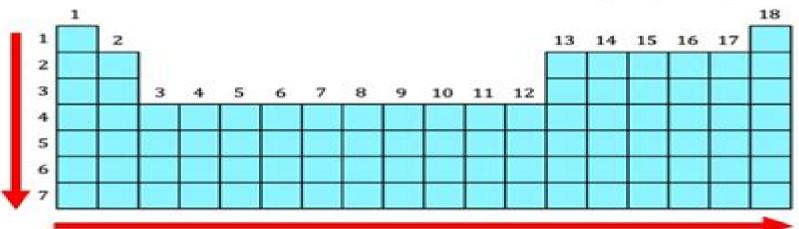
Periodic Trends - Atomic Mass

- The atomic mass describes the mass of each individual element.
- Atomic mass increases moving down each group and right across a period.

<u>Instructions</u>: In the periodic table below, draw two arrows representing the direction that atomic mass <u>increases</u> across groups and periods.



Instructions: In each question, identify the element with the greatest atomic mass:

1) Chlorine (CI), Iodine (I), Bromine (Br), Fluorine (F)

lodine

2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B)

Oxygen

3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe)

Xenon

Instructions: In each question, list the elements from least to greatest atomic mass:

4) Magnesium (Mg), Beryllium (Be), Barium (Ba)

Be, Mg, Ba

5) Gallium (Ga), Selenium (Se), Potassium (K)

K, Ga, Se

Periodic Trends Worksheet Extension Questions

Anthony S. Fauci

Periodic Trends Worksheet Extension Questions:

The Enigmatic Realm of **Periodic Trends Worksheet Extension Questions**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Periodic Trends Worksheet Extension Questions** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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