

Name: _____ () Class: _____ Date: _____

The Periodic Table**General Properties of the Periodic Table**

1. The Periodic Table is a table that groups elements according to certain physical properties or chemical properties.
2. The elements are arranged in order of **increasing proton number (atomic number)**.
3. A **HORIZONTAL** row of elements is known as a **Period**.
Across a period, the element changes from **metal** character to **non-metal** character.
Elements on left hand side of Periodic Table are _____.
Elements on right hand side of Periodic Table are _____.
4. A **VERTICAL** row of elements is known as a **Group**.
5. Hydrogen and Helium are in the same **Period 1**.
They all have _____ **electron shell**.

Draw out the electronic structure of Hydrogen and Helium below:



Hydrogen



Helium

6. The following elements are in the same **Period 2**.
They all have _____

Draw out the electronic structure of these elements.

--	--	--	--	--	--	--	--

Lithium

Beryllium

Boron

Carbon

Nitrogen

Oxygen

Fluorine

Neon

7. Lithium is in **Group 1**. Lithium has _____ outer electron.
Beryllium is in **Group 2**. Beryllium has _____ outer electrons.
Oxygen is in **Group 6**. Oxygen has _____

Quick Check

Without reference to the Periodic Table, predict the Period and Group of the elements below:

- | | |
|-------------------------------|-----------------------------------|
| a) Ca – 2.8.8.2 _____ | d) Xe 2.8.18.18.8 _____ |
| b) S – 2.8.6 _____ | e) Pb 2.8.18.32.18.4 _____ |
| c) Ga – 2.8.18.3 _____ | f) Rb – 2.8.18.8.1 _____ |

Periodic Table The Science Spot Answer Key

Daniela Niemeyer



Periodic Table The Science Spot Answer Key:

Spotlight Science Keith Johnson, Sue Adamson, Gareth Williams, 2002 This Spiral Edition Teacher Support Pack offers comprehensive support and guidance providing the best possible learning experience for your students and saving time for everyone in the department

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Scientifica Essentials, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **The California Electricity Crisis** Charles J. Cicchetti,Jeffrey Dubin,Colin M. Long,2004-06-16 This book attempts to explain what went wrong in California s restructured energy markets and what must be done to restore California s economy and build new electricity systems The intention here is to reconcile the principles of competition and regulation California had a severe electricity crisis for about thirteen months beginning in May of 2000 The economic consequences and political fallout that arose from this crisis persist California s economy continues to suffer and the state s treasury is deeply in debt The state s three investor owned utilities were nearly financially decimated San Diego Gas Electric has recovered to a greater degree than the other two only because its retail prices are about three times the national average and for a time well above the other two IOUs in California Southern California Edison has recently been restored to investment grade and was granted a rate increase Pacific Gas Electric is emerging from bankruptcy This book discusses all of this in greater detail The problems and consequences arising from California s ill fated

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







































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