

CECIL CONVENT SCHOOL

PERIODIC ASSESSMENT-2

Class 11 - Chemistry

Time Allowed: 1 hour

Maximum Marks: 20

General Instructions:

1. There are 11 questions in this question paper.
2. SECTION A - Q. No. 1 to 5 are very short answer questions carrying 1 mark each.
3. SECTION B - Q. No. 6 to 8 are short answer questions carrying 2 marks each.
4. SECTION C- Q. No. 9 to 11 is case based question carrying 3 marks each.
5. All questions are compulsory.
6. Attempt questions in order.

1. Water can be lifted into the water tank at the top of the house with the help of a pump. Then why is it not considered to be spontaneous? [1]
2. Neither q nor W is a state function but $q + W$ is a state function. Explain why? [1]
3. What is the enthalpy change for an adiabatic process? [1]
4. State the law of chemical equilibrium. [1]
5. A tank is full of water. Water is coming in as well as going out at same rate. What will happen to level of water in a tank? What is name given to such state? [1]
6. Calculate the standard enthalpy of formation of $\text{CH}_3\text{OH}(\text{l})$ from the following data: [2]



7. The equilibrium constant for a reaction is 10. What will be the value of ΔG^\ominus ? $R = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$, $T = 300 \text{ K}$. [2]
8. Write the equilibrium constant (K_c) expression for the following reactions. [2]



9. i. Why standard entropy of an elementary substance is not zero whereas standard enthalpy of formation is taken as zero? [3]
 ii. Under what conditions will the reaction occur, if
 a. both ΔH and ΔS are positive
 b. both ΔH and ΔS are negative
10. Dihydrogen gas is obtained from the natural gas by partial oxidation with steam as per following endothermic reaction: [1]
 $\text{CH}_4(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{CO}(\text{g}) + 3\text{H}_2(\text{g})$
 a. Write the expression for K_p for the above reaction.

Periodic Assessment 2 Chemistry Answers

**J.C. Anderson, Keith D. Leaver, Rees D.
Rawlings, Patrick S. Leever**

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Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more [Revise As and A2 - Chemistry](#) Rob Ritchie, 2008-10 Revise AS A2 Chemistry gives complete study support throughout the two A Level years This Study Guide matches the curriculum content and provides in depth course coverage plus invaluable advice on how to get the best results in the exams *Class 9 Chemistry Quiz PDF: Questions and Answers Download | 9th Grade Chemistry Quizzes Book* Arshad Iqbal, The Book Class 9 Chemistry Quiz Questions and Answers PDF Download 9th Grade Chemistry Quiz PDF Book Chemistry Interview Questions for Teachers Freshers Chapter 1 8 Practice Tests Class 9 Chemistry Textbook Questions to Ask in Job Interview includes revision guide for problem solving with hundreds of solved questions Class 9 Chemistry Interview Questions and Answers PDF covers basic concepts analytical and practical assessment tests Class 9 Chemistry Quiz Questions PDF Book helps to practice test questions from exam prep notes The e Book Class 9 Chemistry job assessment tests with answers includes revision guide with verbal quantitative and analytical past papers solved tests Class 9 Chemistry Quiz Questions and Answers PDF Download a book covers solved common questions and answers on chapters Chemical reactivity electrochemistry fundamentals of chemistry periodic table and periodicity physical states of matter solutions structure of atoms structure of molecules tests for school and college revision guide Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The Book Class 9 Chemistry Interview Questions Chapter 1 8 PDF includes high school question papers to review practice tests for exams Class 9 Chemistry Practice Tests a textbook s revision guide with chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam 9th Grade Chemistry Questions Bank Chapter 1 8 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Chemical Reactivity Questions Chapter 2 Electrochemistry Questions Chapter 3 Fundamentals of Chemistry Questions Chapter 4 Periodic Table and Periodicity Questions Chapter 5 Physical States of Matter Questions Chapter 6 Solutions Questions Chapter 7 Structure of Atoms Questions Chapter 8 Structure of Molecules Questions The e Book Chemical Reactivity quiz questions PDF chapter 1 test to download interview questions Metals and non metals The e Book Electrochemistry quiz questions PDF chapter 2 test to download interview questions Corrosion and prevention electrochemical cells electrochemical industries oxidation and reduction oxidation reduction and reactions oxidation states oxidizing and reducing agents The e Book Fundamentals of Chemistry quiz questions PDF chapter 3 test to download interview questions Atomic and mass number Avogadro number and mole branches of chemistry chemical calculations elements and compounds particles elements compounds and mixtures empirical and molecular formulas gram atomic mass molecular mass and gram formula ions and free radicals molecular and formula mass relative atomic mass and mass unit The e Book Periodic Table and Periodicity quiz questions PDF chapter 4 test to download interview questions

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Strengthening Environmental Programs National Science Board (U.S.). Subcommittee on Environmental Programs, 1976

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"Evaluation of Learning" Dr. P.N. Lakshmi Shanmugam, **Journal of Research of the National Bureau of Standards**, 1980 Eye Tracking for

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