

# CHEMICAL REACTIONS

# **Practice Problems**

In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter I.

#### SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

- Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
- Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
- Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
- 4. Write a balanced equation for the production of HCl gas from its elements.
- Write a sentence that completely describes the chemical reaction represented by this balanced equation.

 $2HCl(aq) + CaCO_3(s) \rightarrow CO_2(g) + CaCl_2(aq) + H_2O(l)$ 

Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.

 $2Ag(s) + S(s) \rightarrow Ag_{s}S(s)$ 

- Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
- 8. Write a balanced equation from this word equation.

aqueous silver nitrate + copper metal -- silver metal + aqueous copper nitrate

9. Write a balanced equation for the following word equation.

phosphorus + oxygen → tetraphosphorous decoxide

#### SECTION 11.2 TYPES OF CHEMICAL REACTIONS

- Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
- Write the balanced equation for the reaction that occurs between aluminum and fluorine.
- Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
- Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
- Write the balanced equation for the combustion of propane (C<sub>3</sub>H<sub>8</sub>) to produce carbon dioxide and water vapor.
- Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

# **Pearson Education 11 Chemical Reactions**

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web in the short story gore by sarah ellis the character of amy is able to effectively deceive her brother lucas that there is an alien like creature in the house in order to get her book back soon after their parents go out to do the usual saturday grocery lucas grabs the book from amy s hands and hides inside the washroom

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web feb 15 2015 setting time this was taken place in amy and lucas house specifically the front door washroom living room and their hallway this story takes place on a saturday afternoon the peace keeping forcers are out at mega foods doing the saturday shop ellis 5 initiating incident marcus denouement

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web in numerous ways amy reveals herself to be resourceful throughout the short story gore by sarah ellis her resourcefulness is demonstrated by her imaginative spirit and capability to think outside of the box her intelligence and her willingness to persevere even when faced with obstacles throughout the journey

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web two 2 specific details from the story to support your opinion we get a biased view of lucas because the narration is in 1st person the reader doesn t get to see lucas point of view maybe amy bullies him about academics amy doesn t really have good opinion of her brother lucas is more like an unevolved thugoid

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