Heinemann Chemistry 2 Student Workbook 3e

Chemistry 2 Student Workbook Answers to Worksheets

Teacher Notes

UNIT 3

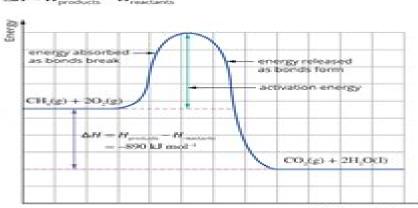
2

Worksheet 01 Crossword—exploring energy

ACROSS		DOWN	
3	Coal seam	1	Galvanic
5	Enthalpy	2	Biofuel
7	Exothermic	4	Spontaneous
11	Cathode	6	Biogas
12	Renewable	8	Joule
14	Thermochemical	9	Oxidation
18	Salt bridge	10	Greenhouse
19	Secondary	13	Activation
21	Hygroscopic	15	Electrolyte
22	Fuel	16	Viscosity
23	Anode	17	Reduction
		20	Fossil
		22	Fuel

Worksheet 02 Combustion of fuels—thermochemical equations and specific heat capacity

The combustion of fuels releases a significant quantity of energy that can be used as an energy source. Chemical energy is the energy stored in the bonds between atoms and between molecules arising from attractions and repulsions between subatomic particles. It is called enthalpy and is represented by the symbol, H. All combustion reactions are exothermic reactions. In such reactions, the enthalpy change, symbol ΔH, has a negative sign. $\Delta H = H_{contacts} - H_{contacts}$



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JE Gale

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