2 Elements, compounds and mixtures

- 1 (a) a substance which cannot be further divided [1] into simpler substances [1] by chemical methods [1]
 - (b) (i) silver, calcium, lead, mercury, zinc, sodium, magnesium [1 for each]
 - (ii) carbon, hydrogen, chlorine, oxygen, astatine, bromine, phosphorus [1 for each]
 - (c) (i) high, good, solids, liquid, malfeable, ductile, lustrous [1 for each]
 - (ii) low, densities, poor, insulators, brittle, gases, dull [1 for each]

Element	Symbol
	Air
Fluorine	
	Cu
Nickel	
Secretary.	Si
Potassium	
	He
Tin	
	BCr .
Aliuminium	

3 (a) substances E and I [1 for each]

(b) substances G and H [1 for each]

(c) substance F [1]

2

(d) substances F and H [1 for each]

(e) substance H [1] (f) substance F [1]

(ii) substained in [1]

(g) substance [[1]

(h) substances E, G and I [1 for each]

(i) substance G [1]

(i) substance [[1]

(a) (i) C [1], C is a non-metal [1]

(ii) iron [1], iron is a metal [1]

(iii) Na [1], its symbol is related to the Latin name [1]

(b) (i) false [1]

GiD true [11]

(Hil) true [1]

(iv) false [1]

(v) true [1]

(vi) false [1]

(viii) true [1]

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University of Cape Town

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