

The number of moles of substances, when compared and reduced to the lowest terms, yields a mole ratio, as shown in the sample below. Indicate the mole ratio for each of the following equations:


2:1:2

1:1:1

4:3:2

1:2:1:1

2:1:2

2:1:1

Convert the following masses to moles by using the information on your periodic table.



$$34.0\text{g} \left(\frac{1\text{mol}}{17.0\text{g}} \right) = 2.00\text{mol}$$

2.00 mol


$$51.4\text{g} \left(\frac{1\text{mol}}{62.0\text{g}} \right) = 0.829\text{mol}$$

0.829 mol


$$4.50\text{g} \left(\frac{1\text{mol}}{98.1\text{g}} \right) = 0.0459\text{mol}$$

0.0459 mol


$$229\text{g} \left(\frac{1\text{mol}}{164.1\text{g}} \right) = 1.40\text{mol}$$

1.40 mol


$$79.5\text{g} \left(\frac{1\text{mol}}{35.0\text{g}} \right) = 2.27\text{mol}$$

2.27 mol

Convert the following to grams.



$$17.1\text{mol} \times \left(\frac{18.0\text{g}}{1\text{mol}} \right) = 308\text{g}$$

308 g.

Pearson Chemistry Answers Stoichiometry

Emilie Sanchez



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IIT Chemistry: Introductory Topics , *The Pearson Guide to Objective Chemistry for the AIEEE* Singhal Atul,2010-09

The Pearson Guide to Physical Chemistry for the IIT JEE Singhal, *Chemistry: The Central Science* Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation The Old Riddle and the Newest Answer John Gerard,1904 **Fundamentals of Physical**

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Experiments in General Organic and Biological Chemistry Robert J. Ouellette, 1988-04-15 *A Text-book of Physics* William Watson, 1904 Data for Science and Technology Phyllis S. Glaeser, 2013-10-22 Data for Science and Technology covers the proceedings of the Seventh International CODATA Conference This text is comprised of 133 chapters with a total of 180 papers from 400 hundred authors which cover CODATA concerned with environmental and energy questions along with problems of data banking and telecommunications network operations This book provides valuable assessment of data and points out alternatives trends and requirements for the future such as production and use of data in pure applied sciences data for the development of human settlements in a dynamic world informatical analysis of scientific research activities and data on our evolutionary heritage Researchers from all scientific fields will find this book a great source reference material since it presents research from various disciplines **The Software Encyclopedia**, 1988

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