

## Chapter 1: Limiting Reagents and Percent Yield (Pages 1-4)

**Remember:** Limiting reagent is the reactant that will completely react and leave another reactant

**Excess:** All reactants except the limiting reagent will be left over. Remember, limiting reagent is the reactant that will completely react.

Remember: Limiting reagent will determine how much product is produced and byproducts will be. The limiting reagent is also a reactant.



The ratio required for the composition for the limiting reagent.

2H <sub>2</sub> (g)	100.00 percent
O <sub>2</sub> (g)	50.00 percent
2H <sub>2</sub> O(l)	100.00 percent
H <sub>2</sub> O(l)	100.00 percent

1. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used. 100.00 g of each reactant will be used.



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# Percent Yield And Limiting Reactants Quiz

**Christian G. Meyer**



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