

# PERIODIC TABLE PUZZLE

Name \_\_\_\_\_

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18

I																		
	F													G	H			
C							E			J			B				A	

	D																

Place the letter of each of the above elements next to its description below.

1. An alkali metal \_\_\_\_\_
2. An alkaline earth metal \_\_\_\_\_
3. An inactive gas \_\_\_\_\_
4. An active nonmetal \_\_\_\_\_
5. A semi-metal \_\_\_\_\_
6. An inner transition element \_\_\_\_\_
7. Its most common oxidation state is -2. \_\_\_\_\_
8. A metal with more than one oxidation state \_\_\_\_\_
9. Metal with an oxidation number of +3 \_\_\_\_\_
10. Has oxidation numbers of +1 and -1 \_\_\_\_\_



# Periodic Table Worksheet Answers Chemistry If8766

**Annelies Wilder-Smith**

## **Periodic Table Worksheet Answers Chemistry If8766:**

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The Basics of the Periodic Table Leon Gray,2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table Included are discussions of the discovery of matter atoms atomic structure molecules compounds ions and isotopes as well as the first identifications of the 118 and counting elements and the various ways they have been classified and organized by prominent scientists up to the present day periodic table Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements their location on the table and their properties and practical uses This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts and furnishes ample opportunities to summarize cite evidence and analyze connections between ideas individuals and events

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**Periodic Table** DK,2020-01-02 A brand new science ebook for kids joins this hugely successful mini encyclopedia series including lot of information Discover all the major elements of the periodic table arranged in chapters according to their group including alkaline earth metals lanthanides and noble gases Every important element from hydrogen via carbon oxygen and gold to oganesson that s element number 118 is presented here Each element is featured in its own catalogue style entry with a photo of the element in its raw state or in use a caption explaining important information ranging from where it s found to its main uses and a fact list presenting the key data including the date of discovery and all atomic information as found on the periodic table Plus there are stunning full page photos showing elements as you have never seen them before perhaps inside machinery that you can t normally delve into or in stunning macro photography that reveals microscopic details invisible to the human eye The style of the Pocket Eyewitness series is perfect for all children from

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R. Scerri, 2019-07-25 The periodic table of elements first encountered by many of us at school provides an arrangement of the chemical elements ordered by their atomic number electron configuration and recurring chemical properties and divided into periodic trends In this Very Short Introduction Eric R Scerri looks at the trends in properties of elements that led to the construction of the table and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and in particular quantum mechanics which underlies the behaviour of all of the elements and their compounds This new edition publishing in the International Year of the Periodic Table celebrates the completion of the seventh period of the table with the ratification and naming of elements 113 115 117 and 118 as nihonium moscovium tennessine and oganesson Eric R Scerri also incorporates new material on recent advances in our understanding of the origin of the elements as well as developments concerning group three of the periodic table **ABOUT THE SERIES** The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable    [The Mathematics of the Periodic Table](#)

D. H. Rouvray, R. Bruce King, 2006 The Periodic Table effectively embraces the whole realm of chemistry within the confines of one comparatively simple and easily understood chart of the chemical elements Over many years the Periodic Table has proven to be indispensable not only to chemists of all kinds but also to a host of other scientists including biologists geologists and physicists It is thus hardly surprising that the Periodic Table has become one of our most celebrated contemporary scientific icons In the present work various aspects of the Periodic Table that are seldom if ever featured elsewhere are given prominence The twelve presentations contained herein all have a mathematical flavour because it is the intention to highlight the often neglected mathematical features of the Periodic Table and several closely related topics The book starts out by considering predictions of what the ultimate size of the Periodic Table will be when all of the possible artificial chemical elements have been synthesised It then moves on to an examination of the nature of the periodicity extant in the Periodic Table and some methods for the prediction of the properties of the super heavy elements The Periodic Table is next explored in various dimensions other than two The natural clustering of the elements into groups is studied by three different but complementary routes namely via the topological structures of the groups the self association of the elements as evidenced by neural network studies and information theoretical analysis of the behaviour of atoms Following a detailed investigation of the mathematical basis for the periodicity seen in atomic and molecular spectroscopy three separate presentations delve into many different aspects of the group theoretical structure of the Periodic Table The unusual combination of themes offered here will appeal to all who seek a more detailed and intimate knowledge of the Periodic Table than that available in standard texts on the subject    [Mastering the Periodic Table](#) Linda Trombley, J. Weston Walch

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