

Periodic Table of the Elements

KEY

Atomic Mass → 12.011 ← Selected Oxidation States
Symbol → C
Atomic Number → 6
Electron Configuration → 2-4

Relative atomic masses are based on $^{12}\text{C} = 12$ (exact)
Notes: Numbers in parentheses are mass numbers of the most stable or common isotope.

1	2											13	14	15	16	17	18
1 1 H												5 11 B	6 12 C	7 14 N	8 16 O	9 19 F	10 20 Ne
Group		Group										Group					
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
2 3 Li	4 9 Be											13 27 Al	14 28 Si	15 31 P	16 32 S	17 35 Cl	18 36 Ar
3 11 Na	12 24 Mg											13 27 Al	14 28 Si	15 31 P	16 32 S	17 35 Cl	18 36 Ar
4 19 K	20 40 Ca	21 45 Sc	22 48 Ti	23 50 V	24 52 Cr	25 55 Mn	26 56 Fe	27 59 Co	28 59 Ni	29 64 Cu	30 65 Zn	31 70 Ga	32 73 Ge	33 75 As	34 79 Se	35 79 Br	36 84 Kr
5 37 Rb	38 88 Sr	39 90 Y	40 91 Zr	41 93 Nb	42 96 Mo	43 98 Tc	44 101 Ru	45 103 Rh	46 106 Pd	47 108 Ag	48 112 Cd	49 115 In	50 117 Sn	51 120 Sb	52 128 Te	53 127 I	54 131 Xe
6 55 Cs	56 137 Ba	57 139 La	58 140 Ce	59 141 Pr	60 143 Nd	61 145 Pm	62 150 Sm	63 152 Eu	64 157 Gd	65 159 Tb	66 163 Dy	67 165 Ho	68 167 Er	69 169 Tm	70 173 Yb	71 175 Lu	
7 87 Fr	88 226 Ra	89 227 Ac	90 228 Th	91 232 Pa	92 238 U	93 237 Np	94 244 Pu	95 247 Am	96 251 Cm	97 252 Bk	98 258 Cf	99 261 Es	100 265 Fm	101 269 Md	102 271 No	103 270 Lr	

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2 3 Li	4 4 Be											5 5 B	6 6 C	7 7 N	8 8 O	9 9 F	10 10 Ne
3 11 Na	12 12 Mg											13 13 Al	14 14 Si	15 15 P	16 16 S	17 17 Cl	18 18 Ar
4 19 K	20 20 Ca	21 21 Sc	22 22 Ti	23 23 V	24 24 Cr	25 25 Mn	26 26 Fe	27 27 Co	28 28 Ni	29 29 Cu	30 30 Zn	31 31 Ga	32 32 Ge	33 33 As	34 34 Se	35 35 Br	36 36 Kr
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Periodic Table Reference Sheet

S Ben Porath



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