

# Periodic Trends – Atomic Mass

- The **atomic mass** describes the mass of each individual element.
- **Atomic mass** increases moving **down** each group and **right** across a period.

**Instructions:** In the periodic table below, draw two arrows representing the direction that atomic mass **increases** across groups and periods.

	1																	18	
1		2												13	14	15	16	17	
2																			
3			3	4	5	6	7	8	9	10	11	12							
4																			
5																			
6																			
7																			

**Instructions:** In each question, identify the element with the **greatest** atomic mass:

- 1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F) Iodine
- 2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B) Oxygen
- 3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe) Xenon

**Instructions:** In each question, list the elements from **least** to **greatest** atomic mass:

- 4) Magnesium (Mg), Beryllium (Be), Barium (Ba) Be, Mg, Ba
- 5) Gallium (Ga), Selenium (Se), Potassium (K) K, Ga, Se

# Periodic Trends Worksheet Extension Questions

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**Periodic Trends Worksheet Extension Questions:**

## **Periodic Trends Worksheet Extension Questions** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Periodic Trends Worksheet Extension Questions**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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