

Periodic Table of the Elements

1 IA H Hydrogen (1.008)																	18 VIIIA He Helium (4.003)		
3 IIA Li Lithium (6.941)	4 IIA Be Beryllium (9.012)													5 IIIA B Boron (10.811)	6 IIIA C Carbon (12.011)	7 IVA N Nitrogen (14.007)	8 IVA O Oxygen (15.999)	9 VA F Fluorine (18.998)	10 VA Ne Neon (20.180)
11 IA Na Sodium (22.990)	12 IIA Mg Magnesium (24.305)													13 IIIA Al Aluminum (26.982)	14 IIIA Si Silicon (28.086)	15 IVA P Phosphorus (30.974)	16 IVA S Sulfur (32.065)	17 VA Cl Chlorine (35.453)	18 VA Ar Argon (39.948)
19 IA K Potassium (39.098)	20 IIA Ca Calcium (40.078)	21 IIB Sc Scandium (44.956)	22 IIB Ti Titanium (47.883)	23 IIB V Vanadium (50.942)	24 IIB Cr Chromium (51.996)	25 IIB Mn Manganese (54.938)	26 IIB Fe Iron (55.845)	27 IIB Co Cobalt (58.933)	28 IIB Ni Nickel (58.693)	29 IB Cu Copper (63.546)	30 IB Zn Zinc (65.38)	31 IIB Ga Gallium (69.723)	32 IIB Ge Germanium (72.64)	33 IIIA As Arsenic (74.922)	34 IIIA Se Selenium (78.96)	35 IIIA Br Bromine (79.904)	36 IIIA Kr Krypton (83.798)		
37 IA Rb Rubidium (85.468)	38 IIA Sr Strontium (87.62)	39 IIB Y Yttrium (88.906)	40 IIB Zr Zirconium (91.224)	41 IIB Nb Niobium (92.906)	42 IIB Mo Molybdenum (95.94)	43 IIB Tc Technetium (98.906)	44 IIB Ru Ruthenium (101.07)	45 IIB Rh Rhodium (102.91)	46 IIB Pd Palladium (106.90)	47 IB Ag Silver (107.87)	48 IB Cd Cadmium (112.41)	49 IIB In Indium (114.82)	50 IIB Sn Tin (118.71)	51 IIIA Sb Antimony (121.76)	52 IIIA Te Tellurium (127.6)	53 IIIA I Iodine (126.91)	54 IIIA Xe Xenon (131.29)		
55 IA Cs Cesium (132.91)	56 IIA Ba Barium (137.33)	57-71 Lanthanides	72 IIB Hf Hafnium (178.49)	73 IIB Ta Tantalum (180.95)	74 IIB W Tungsten (183.84)	75 IIB Re Rhenium (186.21)	76 IIB Os Osmium (190.23)	77 IIB Ir Iridium (192.22)	78 IIB Pt Platinum (195.08)	79 IB Au Gold (196.97)	80 IB Hg Mercury (200.59)	81 IIB Tl Thallium (204.38)	82 IIB Pb Lead (207.2)	83 IIIA Bi Bismuth (208.98)	84 IIIA Po Polonium (209)	85 IIIA At Astatine (210)	86 IIIA Rn Radon (222)		
87 IA Fr Francium (223)	88 IIA Ra Radium (226)	89-103 Actinides	104 IIB Rf Rutherfordium (261)	105 IIB Db Dubnium (262)	106 IIB Sg Seaborgium (266)	107 IIB Bh Bohrium (264)	108 IIB Hs Hassium (277)	109 IIB Mt Meitnerium (268)	110 IIB Ds Darmstadtium (271)	111 IB Rg Roentgenium (272)	112 IB Cn Copernicium (285)	113 IIB Nh Nihonium (284)	114 IIB Fl Flerovium (289)	115 IIIA Mc Moscovium (288)	116 IIIA Lv Livermorium (293)	117 IIIA Ts Tennessine (289)	118 IIIA Og Oganesson (294)		

Atomic Number →

→ Symbol

→ Name

→ Atomic Weight

→ Electrons per shell

1

H

Hydrogen

1.008

1

57 La Lanthanum (138.905)	58 Ce Cerium (140.12)	59 Pr Praseodymium (140.908)	60 Nd Neodymium (144.24)	61 Pm Promethium (145)	62 Sm Samarium (150.36)	63 Eu Europium (151.964)	64 Gd Gadolinium (157.25)	65 Tb Terbium (158.925)	66 Dy Dysprosium (162.50)	67 Ho Holmium (164.930)	68 Er Erbium (167.259)	69 Tm Thulium (168.933)	70 Yb Ytterbium (173.054)	71 Lu Lutetium (174.967)
89 Ac Actinium (227)	90 Th Thorium (232.038)	91 Pa Protactinium (231.036)	92 U Uranium (238.029)	93 Np Neptunium (237.048)	94 Pu Plutonium (244.064)	95 Am Americium (243.061)	96 Cm Curium (247.070)	97 Bk Berkelium (247.070)	98 Cf Californium (251.083)	99 Es Einsteinium (252.083)	100 Fm Fermium (257.103)	101 Md Mendelevium (258.103)	102 No Nobelium (259.103)	103 Lr Lawrencium (262.103)

Periodic Table Section 3

Julia Schneider



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Structuralist Knowledge Representation , 2022-05-16

Organic Chemistry Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps

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