

2.6 - Periodic Trends worksheet

Name _____

I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

II. Answer the questions using the symbols on the periodic table

1. What elements are in period 5?

What element(s) is a

- | | |
|-------------------------------|---------------------------|
| _____ 2.halogens | _____ 6.transition metals |
| _____ 3. lanthanides | _____ 7.alkali metals |
| _____ 4.alkaline earth metals | _____ 8.actinides |
| _____ 5.noble gases | |

What elements are in the sublevel

- _____ 9. s
_____ 10. p
- _____ 11. d
_____ 12. f

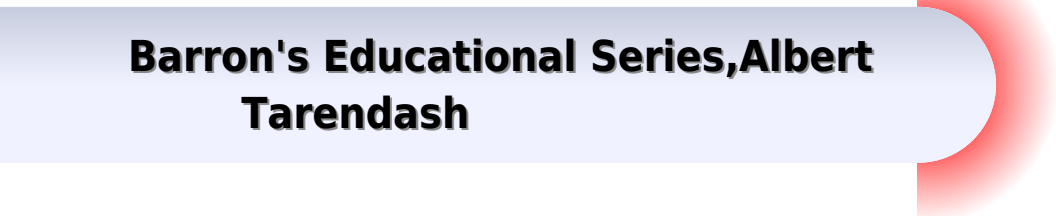
What elements are

13. metals.
14. nonmetals
15. metalloids,
16. Which element is the most reactive nonmetal?
17. Which element is the most reactive metal?
18. Which element below is the least reactive?
19. Which element has the largest radius in period 4?
20. Which element has the largest radius in the Alkali metals family?
21. Which element has the highest ionization energy in period four?
22. Which element has the second largest atomic number?
23. Which element has the largest atomic mass?
24. Which element is located at period 4, group 15?
25. Which element is located at period 5, group 2?
27. Which elements will form +1 ions?
28. Which group of metals will have the same properties
29. Which group of nonmetals will have the same properties
30. On the back, define all chapter 5, section 3 vocabulary terms.

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Periodic Table Chemistry Answers

**Barron's Educational Series, Albert
Tarendash**



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1. Understanding the eBook Periodic Table Chemistry Answers
 - The Rise of Digital Reading Periodic Table Chemistry Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Chemistry Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Chemistry Answers
 - Personalized Recommendations
 - Periodic Table Chemistry Answers User Reviews and Ratings
 - Periodic Table Chemistry Answers and Bestseller Lists
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 - Highlighting and Note-Taking Periodic Table Chemistry Answers
 - Interactive Elements Periodic Table Chemistry Answers
- 8. Staying Engaged with Periodic Table Chemistry Answers
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Chemistry Answers
- 9. Balancing eBooks and Physical Books Periodic Table Chemistry Answers
 - Benefits of a Digital Library
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- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Chemistry Answers
 - Setting Reading Goals Periodic Table Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Chemistry Answers
 - Fact-Checking eBook Content of Periodic Table Chemistry Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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