

Mars Periodic Table

Use the following clues to put the elements in their proper places on the periodic table. These symbols have absolutely no relationship to our symbols. ("O", for example, does not necessarily represent Oxygen here).

The symbols are: A, B, C, D, E, F, G, H, I, J, K, L, M, N, O, P, Q, R, S, T, U, V, W, X, Y, Z, @, #, \$, %.

The most reactive metal is X.

The lightest halogen is I.

The noble gases are B, L, W and J. B is the lightest of these, W is the heaviest of these and L is in period 2.

The lightest element on the periodic table is A.

The following elements all have three electron shells: %, H, K, M, O, Y and Z. % is an alkali metal. H can form +4 or -4 charges. K belongs in the same family as I. M forms -3 charges. O forms -2 charges. Y is a metalloid by location but actually acts more like a light metal. Z is an alkaline earth metal.

G has 7 total electrons but only 5 valence electrons.

C has an average atomic mass of approximately 7 amu.

Q has 4 electron shells and only 1 electron in that outer shell.

The E family is made up of the elements E, #, R, and Y. E is a metalloid. # is in the same period as X.

\$ has the greatest atomic mass of all the elements on the periodic table.

P is in period 5 and D is in period 2. They are both in the same family as Z.

F has an atomic number of 6. It is in the same family as H, T and \$

When saltwater is distilled (salt is removed), it produces the liquid pure A_2N .

V is a halogen. @@ is an alkaline earth metal. S is a metalloid. U is in period 4.

1	2	13	14	15	16	17	18
A							B
C	D	E	F	G	N	I	L
%	Z	Y	H	M	O	K	J
Q	@	R	T	S	U	V	W
X	P	#	\$				

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Mike Jess



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Table of Contents Periodic Law Puzzle Answers

1. Understanding the eBook Periodic Law Puzzle Answers
 - The Rise of Digital Reading Periodic Law Puzzle Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Law Puzzle Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Law Puzzle Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Law Puzzle Answers
 - Personalized Recommendations

- Periodic Law Puzzle Answers User Reviews and Ratings
- Periodic Law Puzzle Answers and Bestseller Lists
- 5. Accessing Periodic Law Puzzle Answers Free and Paid eBooks
 - Periodic Law Puzzle Answers Public Domain eBooks
 - Periodic Law Puzzle Answers eBook Subscription Services
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- 6. Navigating Periodic Law Puzzle Answers eBook Formats
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 - Periodic Law Puzzle Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Law Puzzle Answers
 - Highlighting and Note-Taking Periodic Law Puzzle Answers
 - Interactive Elements Periodic Law Puzzle Answers
- 8. Staying Engaged with Periodic Law Puzzle Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Law Puzzle Answers
- 9. Balancing eBooks and Physical Books Periodic Law Puzzle Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Law Puzzle Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Law Puzzle Answers
 - Setting Reading Goals Periodic Law Puzzle Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Law Puzzle Answers
 - Fact-Checking eBook Content of Periodic Law Puzzle Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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