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 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $3\text{HBr} + 1\text{KHCO}_3 \rightarrow 1\text{H}_2\text{O} + 1\text{KBr} + 1\text{CO}_2$
 $4\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow 1\text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 5\text{S} + 3\text{H}_2\text{O}$
 $2\text{NF}_3 + 3\text{SnF}_2 \rightarrow 1\text{N}_2\text{O}_3 + 2\text{NaOH} + 2\text{H}_2\text{SO}_4$
 balance using the following equation

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 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $4\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 5\text{S} + 3\text{H}_2\text{O}$
 $2\text{NF}_3 + 3\text{SnF}_2 \rightarrow \text{N}_2\text{O}_3 + 2\text{NaOH} + 2\text{H}_2\text{SO}_4$

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 $\text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$

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 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $4\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3\text{NaBr} + 5\text{S} + 3\text{H}_2\text{O}$
 $2\text{NF}_3 + 3\text{SnF}_2 \rightarrow \text{N}_2\text{O}_3 + 2\text{NaOH} + 2\text{H}_2\text{SO}_4$
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