



SA MILITARY HEALTH SERVICE



APPLICATION FOR MILITARY SKILLS DEVELOPMENT (MSDS) - 2024 INTAKE

COMPLETE THE FOLLOWING:

Biographical information

First Name										
Surname										
Birth Country										
Citizenship										
ID No										
Your Cell No										
Alternative 1										
Alternative 2										
Postal Address										
Town/City										
Province										
Postal code										
I will be			years of age on 8 January 2024							

Province, gender and race (for statistical purposes, please select one)

Province: ☐ Gauteng ☐ Eastern Cape ☐ Free State
☐ Northern Cape ☐ Mpumalanga ☐ Western Cape
☐ North West ☐ KwaZulu Natal (KZN) ☐ Limpopo

Gender: ☐ Male ☐ Female
 Race: ☐ African ☐ Asian
☐ Coloured ☐ White

Tertiary Qualifications

☐ N6 ☐ 3 Year Diploma ☐ Degree

Attach your CV and Copies of: ID, Matric Certificate (only statement of results for 2022 will be accepted) and Tertiary Qualifications (after confirming completion of qualification will be accepted)

Other Requirements

Applicants will be subjected to drug testing, fitness, Psychometric and medical evaluation; verification of Criminal record, prescreening and verification of Qualifications.

Related Enquiries: Tel: 012 671 5053/5195/5168/ 5002

MSDS BACKGROUND

The South African Military Health Service (SAMHS), through its Military Skills Development System (MSDS), is offering young South African citizens an opportunity to serve in uniform on a two-year contract. The Recruits will receive military training and further functional training in their first year of service. During the second year of service, depending on the duration of the functional orientation, they will be deployed where needed and given opportunity to apply their knowledge and develop their skills.

General Requirements:

- RSA Citizen (No dual citizenship).
- 18 – 22 years of age (Gr 12).
- 18 - 26 (3 years Tertiary Qualification) when reporting for the January 2024 Intake.
- Admission Point System (APS) Score of 24, excluding Life Orientation (LO).
- Have no criminal record or pending cases.
- Comply with medical fitness requirements for the appointment.
- No Visible tattoos (Legs, Lower Arm, Neck and Face).
- Physical Fitness Test will be conducted during selections.

Select one of the following fields of interest:

- ☐ Patient Administration. ☐ Logistics. ☐ Medical Support Operations. ☐ Dog Handler. ☐ Counter Intelligence (Gr 12 with ALL the following Subjects: English Level 4, 2nd Language Level 4, Mathematics/Literacy Level 3, Life Science Level 3 and Geography/Physical Science Level 3). ☐ Corporate Communication (Gr 12 plus a Degree in Communication/Diploma in Public Relations and Journalism). ☐ Human Resource (Gr 12 plus Bachelors Degree in Organisational Study). ☐ Chemical Biological Radiation Defence (Gr 12 with ALL Subjects on Level 5: English, Mathematics (NO Maths Lit), Life Science and Physical Science). ☐ Environmental Health (Gr 12 with All the following Subjects: Mathematics (NO Maths Lit) Level 5, Physical Science (Level 5) and Life Science Level 4).
- ☐ Musician (Gr 12 plus a Music Practical Gr 6 or higher Certificate from a recognised institution (UNISA, TRINITY, ROYAL SCHOOLS).
- ☐ Dental Assistant (Gr 12 plus a Dental Assistance Certificate).
- ☐ Nursing Diploma (Gr 12 with ALL Subjects on Level 4: English, Life Science and Mathematics/Literacy). ☐ Auxiliary Nursing (Gr 12 with ALL Subjects on Level 3: English, Life Science and Mathematics/Literacy). ☐ Emergency Care Assistant (Gr 12 with ALL Subjects: English Level 5, Mathematics Level 3 (NO Maths Lit), Physical Science Level 4 and Life Science Level 3).

Hand deliver your application to the nearest recruitment centre or via post at the SAMHS postal address:

**SAMHS Recruitment Office – Department of Defence
Private Bag X102
Centurion, Pretoria, 0046**

Signature of Applicant

Date

If you have not been contacted by 31 December 2023, kindly consider your application to be unsuccessful.

MSDS APPLICATION CLOSING DATE: 28 February 2023

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Samhs Application Forms 2015

Hussin A.Rothana



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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$ one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $H_2C_2O_4 \cdot 2H_2O$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web jul 19 2023 introduction titration is an analytical quantitative technique used to determine the concentration of a solute a pH titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pKa of an unknown acid or pKb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ $4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web conclusion in this lab we used titration to explore the concepts of stoichiometry and equivalence points we found the concentration of an unknown substance by mixing 2M HCl with the NaOH of unknown concentration in order to experimentally ascertain the concentration of the NaOH

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the