

TABLE 3: THE PERIODIC TABLE OF ELEMENTS  
TABEL 3: DIE PERIODIEKE TABEL VAN ELEMENTE

| 1<br>(I)        | 2<br>(II)       | 3               | 4               | 5               | 6               | 7               | 8               | 9               | 10              | 11               | 12              | 13<br>(III)     | 14<br>(IV)       | 15<br>(V)        | 16<br>(VI)       | 17<br>(VII)      | 18<br>(VIII)    |
|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|-----------------|------------------|-----------------|-----------------|------------------|------------------|------------------|------------------|-----------------|
| 1<br>H<br>1     |                 |                 |                 |                 |                 |                 |                 |                 |                 |                  |                 |                 |                  |                  |                  |                  | 2<br>He<br>4    |
| 3<br>Li<br>7    | 4<br>Be<br>9    |                 |                 |                 |                 |                 |                 |                 |                 |                  |                 |                 |                  |                  |                  |                  |                 |
| 11<br>Na<br>23  | 12<br>Mg<br>24  |                 |                 |                 |                 |                 |                 |                 |                 |                  |                 |                 |                  |                  |                  |                  |                 |
| 19<br>K<br>39   | 20<br>Ca<br>40  | 21<br>Sc<br>45  | 22<br>Ti<br>48  | 23<br>V<br>51   | 24<br>Cr<br>52  | 25<br>Mn<br>55  | 26<br>Fe<br>56  | 27<br>Co<br>59  | 28<br>Ni<br>59  | 29<br>Cu<br>63,5 | 30<br>Zn<br>65  | 31<br>Ga<br>70  | 32<br>Ge<br>73   | 33<br>As<br>75   | 34<br>Se<br>79   | 35<br>Br<br>80   | 36<br>Kr<br>84  |
| 37<br>Rb<br>86  | 38<br>Sr<br>88  | 39<br>Y<br>89   | 40<br>Zr<br>91  | 41<br>Nb<br>92  | 42<br>Mo<br>96  | 43<br>Tc<br>96  | 44<br>Ru<br>101 | 45<br>Rh<br>103 | 46<br>Pd<br>106 | 47<br>Ag<br>108  | 48<br>Cd<br>112 | 49<br>In<br>115 | 50<br>Sn<br>119  | 51<br>Sb<br>122  | 52<br>Te<br>128  | 53<br>I<br>127   | 54<br>Xe<br>131 |
| 55<br>Cs<br>133 | 56<br>Ba<br>137 | 57<br>La<br>139 | 72<br>Hf<br>179 | 73<br>Ta<br>181 | 74<br>W<br>184  | 75<br>Re<br>186 | 76<br>Os<br>190 | 77<br>Ir<br>192 | 78<br>Pt<br>195 | 79<br>Au<br>197  | 80<br>Hg<br>201 | 81<br>Tl<br>204 | 82<br>Pb<br>207  | 83<br>Bi<br>209  | 84<br>Po<br>209  | 85<br>At<br>210  | 86<br>Rn<br>222 |
| 87<br>Fr<br>223 | 88<br>Ra<br>226 | 89<br>Ac        |                 |                 |                 |                 |                 |                 |                 |                  |                 |                 |                  |                  |                  |                  |                 |
|                 |                 |                 | 58<br>Ce<br>140 | 59<br>Pr<br>141 | 60<br>Nd<br>144 | 61<br>Pm<br>144 | 62<br>Sm<br>150 | 63<br>Eu<br>152 | 64<br>Gd<br>157 | 65<br>Tb<br>159  | 66<br>Dy<br>163 | 67<br>Ho<br>165 | 68<br>Er<br>167  | 69<br>Tm<br>169  | 70<br>Yb<br>173  | 71<br>Lu<br>175  |                 |
|                 |                 |                 | 90<br>Th<br>232 | 91<br>Pa<br>231 | 92<br>U<br>238  | 93<br>Np<br>237 | 94<br>Pu<br>244 | 95<br>Am<br>243 | 96<br>Cm<br>247 | 97<br>Bk<br>247  | 98<br>Cf<br>251 | 99<br>Es<br>252 | 100<br>Fm<br>257 | 101<br>Md<br>288 | 102<br>No<br>289 | 103<br>Lr<br>260 |                 |

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