Blackline Master 5.8

Ionic Compounds: Names and Formulas Worksheet

(a) magnesium oxide		MgO	(k) copper(l) bromide		CuBr
(b) sodium fluoride		NaP			Sn Brz
(c) aluminum nitride		AIN	(m) iron(III) chloride		Fects
(d) potassium sulfide		_ K25	(n) calcium phosphide		Car Pa
(e) lithium iodide		LiT	(o) lead(II) oxide		Pbo
(f) calcium bromide		Cabra	(p) lead(IV) fluoride		Pb Fy
(g) beryllium oxide		Beo	(q) tin(IV) bromide		Sn Bry
(h) nickel chloride		Nio	(r) copper(II) sulfide		Cus
(i) magnesium nitride		MgzNz	(s) iron(II) axide		Feo
(j) aluminum sulfide			(t) calcium nitride		
Write the	names for the f	AlsS3			Ca 3 N2
Write the	names for the f	ollowing compounds.	(k) PbS	lead	(III) su(ficle
Write the a) Li ₂ O b) AlCl ₃	lithium aluminu	ollowing compounds. oxide un chloride	(k) PbS _ (l) SnO ₂ _	lead tin (D	(II) su(ficle I) oxide
Write the le) Li ₂ O b) AICl ₃ c) MgS	lithium aluminu magnes	ollowing compounds. oxide un chloride ium sulfide	(k) PbS (l) SnO ₂ (m) Na ₂ S	lead tin (I) sodium	(II) su(ficle Z) oxide sulficle
Write the a) Li ₃ O b) AlCl ₃ c) MgS d) CaO	names for the f lithium aluminu magnes calcius	ollowing compounds. oxide un chloride ium sulfide m oxide	(k) PbS (l) SnO ₂ (m) Na ₂ S (n) Mg ₃ P ₂	lead tin (I sudium magnes	(II) su(ficle Z) oxide su(ficle jum phosphid
Write the la) Li ₃ O b) AlCl ₃ c) MgS d) CaO	names for the f lithicum aluminu magnes calcius	ollowing compounds. oxide un chloride ium sulfide m oxide un bromide	(k) PbS (l) SnO ₂ (m) Na ₂ S (n) Mg ₂ P ₂ (o) NiO	lead fin (I) sudium magnes nickel	(II) su(ficle I) oxide su(ficle jum phosphid (II) oxide
Write the (a) Li ₃ O (b) AlCl ₃ (c) MgS (d) CaO (e) KBr	names for the f lithicum aluminu magnes calcius potassi beryllic	ollowing compounds. oxide un chloride ium sulfide moxide un bromide um fluoride	(k) PbS	lead tin (I) sudium magnes nickel copper	(II) su(ficle I) oxide sulficle ium phosphid (II) oxide (II) oxide
Write the (a) Li ₃ O (b) AlCl ₃ (c) MgS (d) CaO (e) KBr	names for the f lithium aluminu magnes calcius potassi beryllic	ollowing compounds. oxide un chloride ium sulfide m oxide un bromide um bromide um fluoride n ni fride	(k) PbS (l) SnO ₂ (m) Na ₂ S (n) Mg ₂ P ₂ (o) NiO	lead fin (I) sodium magnes nickel copper lead ((II) sulficle Sulficle sulficle ium phosphid (II) oxide (II) iodide (II) chloride
Write the a) Li ₂ O b) AlCl ₃ c) MgS d) CaO e) KBr f) BeF	magnes aluminu aluminu magnes calcius potassi benylliu sodiuv aluminu	ollowing compounds. oxide un chloride ium sulfide m oxide un bromide um fluoride n ni tride um oxide	(k) PbS	lead fin (I) sudium magnes nickel copper lead (iron ()	(II) su(ficle I) oxide sulficle jum phosphid (II) oxide (II) oxide (II) indide (II) chloride (III) phosphide
Write the	names for the 1 lithium aluminu magnes calcius potassi beryllic sodius copper (1	ollowing compounds. oxide un chloride ium sulfide m oxide un bromide um bromide um fluoride n ni fride	(k) PbS	lead fin (I) sudium magnes nickel copper lead (iron () calciu	(II) sulficle Sulficle sulficle ium phosphid (II) oxide (II) iodide (II) chloride

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Christian G. Meyer

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