



## GRADE 10 PHYSICAL SCIENCES

### NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

#### SECTION A

##### QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write ONLY the LETTER (A-D) next to the QUESTION/NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration  $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B  $Na_2S$
- C  $NaS_2$
- D  $Na_2S_2$

1.4 Strongest intermolecular force found in water?

- A ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed
- B vibrate against each other
- C move freely and rarely touch
- D move over one another

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a

|             |    |
|-------------|----|
| $1s^2$      | 2s |
| $2s^2 2p^4$ | 3s |
| $3s^2$      | 3p |
| $3d^5$      | 4s |

- A noble gas
- B halogen
- C alkali earth metal
- D alkali metal

# Physical Science November Grpaper2 2013

**O García**



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### **Table of Contents Physical Science November Grpaper2 2013**

1. Understanding the eBook Physical Science November Grpaper2 2013
  - The Rise of Digital Reading Physical Science November Grpaper2 2013
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science November Grpaper2 2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science November Grpaper2 2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science November Grpaper2 2013
  - Personalized Recommendations
  - Physical Science November Grpaper2 2013 User Reviews and Ratings
  - Physical Science November Grpaper2 2013 and Bestseller Lists

5. Accessing Physical Science November Grpaper2 2013 Free and Paid eBooks
  - Physical Science November Grpaper2 2013 Public Domain eBooks
  - Physical Science November Grpaper2 2013 eBook Subscription Services
  - Physical Science November Grpaper2 2013 Budget-Friendly Options
6. Navigating Physical Science November Grpaper2 2013 eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science November Grpaper2 2013 Compatibility with Devices
  - Physical Science November Grpaper2 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science November Grpaper2 2013
  - Highlighting and Note-Taking Physical Science November Grpaper2 2013
  - Interactive Elements Physical Science November Grpaper2 2013
8. Staying Engaged with Physical Science November Grpaper2 2013
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science November Grpaper2 2013
9. Balancing eBooks and Physical Books Physical Science November Grpaper2 2013
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science November Grpaper2 2013
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science November Grpaper2 2013
  - Setting Reading Goals Physical Science November Grpaper2 2013
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science November Grpaper2 2013
  - Fact-Checking eBook Content of Physical Science November Grpaper2 2013
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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