



MERCY JUNIOR COLLEGE
Grade 11
Physical Science
Midterm Exam

NAME: _____
YEAR LEVEL & SECTION: _____
TEACHER'S NAME: _____
PROCTOR'S NAME: _____

I. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol e^- or β^- , with a negative elementary electric charge.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol p or p^+ , with a positive electric charge of $+1e$ elementary charge and mass slightly less than that of a neutron.
 - Neutron
 - Electron
 - Proton
 - Atom
- It is a subatomic particle, symbol n or n^0 , with no net electric charge and a mass slightly larger than that of a proton.
 - Neutron
 - Electron
 - Proton
 - Atom
- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

- It is a type of radioactive decay in which an atomic nucleus emits an alpha particle and thereby transforms or 'decays' into an atom with a mass number that is reduced by four and an atomic number that is reduced by two.
 - Alpha Decay
 - Beta Decay
 - Gamma Ray
 - Radiation

- It is a type of radioactive decay in which a beta ray (fast energetic electron or positron) and a neutrino are emitted from an atomic nucleus.

- Alpha Decay
- Beta Decay
- Gamma Ray
- Radiation

- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.

- Electron Configuration
- Periodic Table of Elements
- Orbital Models
- Energy Levels

- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

Physical Science 2june Exam Paper 1

Egbert Boeker, Rienk van Grondelle



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