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4.14.

$$\left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \quad \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

1000 g of water contains 1000 g of water.

- 4.15. (a) The water molecule has two lone pairs of electrons on each oxygen.
 (b) The water molecule has two lone pairs of electrons on each oxygen.
 (c) The water molecule has two lone pairs of electrons on each oxygen.

4.16. (a) The water molecule has two lone pairs of electrons on each oxygen.
 (b) The water molecule has two lone pairs of electrons on each oxygen.

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \quad \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

4.17. (a) The water molecule has two lone pairs of electrons on each oxygen.

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \quad \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \quad \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

4.18. (a) The water molecule has two lone pairs of electrons on each oxygen.

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \quad \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

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Katrin Zwirglmaier



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